Category

Polymer-Supported Synthesis

Key Words

cooperative
phenomena
heterogeneous
catalysis
mesoporous
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hybrid composites
silicon

S. HUN, H.-T. CHEN, J. W. WIENCH, M. PRUSKI, V. S.-Y. LIN* (IOWA STATE UNIVERSITY, USA) Cooperative Catalysis by General Acid and Base Bifunctionalized Mesoporous Silica Nanospheres *Angew. Chem. Int. Ed.* **2005**, *44*, 1826-1830.

Cooperative Catalysis by General Acid and Base Bifunctionalized Mesoporous Silica Nanospheres

O₂N
$$O_2$$
N O_2 N O_2 N O_2 N O_2 N O_2 N O_2 N O_3 N O_4 N O_5 N O

Significance: Bifunctinalized mesoporous silica nanospheres (MSNs) having a general acid, the ureidopropyl group (UDP), and a general base, 3-[2-(2-aminoethylamino)ethylamino]propyl group (AEP), were synthesized. The AEP/UDP-MSNs catalyzed the carbonyl activation reactions, aldol reaction, Henry reaction, and cyanosilylation with high TONs up to 22.6, 125.0, and 276.1, respectively.

Comment: Recently, a new synthetic method to functionalize mesoporous silica with multiple organic groups was reported from the author's group, by which the loading of functional groups, particle size, and shape of the mesoporous silicas could be controlled (for example, see: S. Huh, J. W. Wiench, J.-C. Yoo, M. Pruski, V. S.-Y. Lin *Chem. Mater.* **2003**, *15*, 4247-4256). The authors discussed that the AEP and the UDP activate both the nucleophile and the electrophile at the same time, cooperatively.

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