Category

Synthesis

Key words

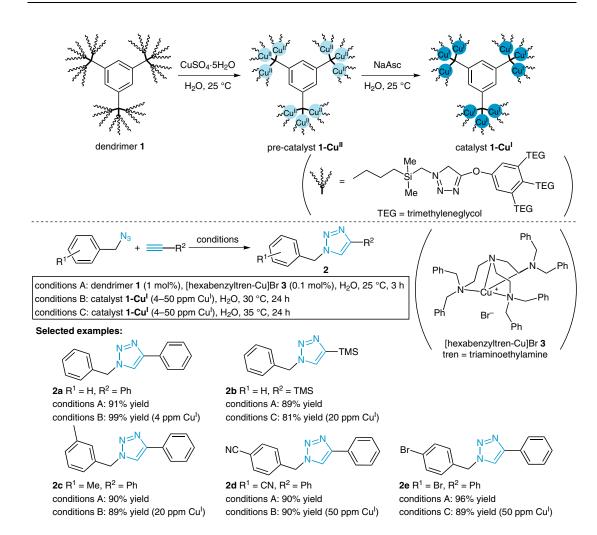
dendrimers

Huisgen cycloaddition click chemistry copper catalysis

Polymer-Supported

C. DERAEDT, N. PINAUD, D. ASTRUC* (UNIVERSITÉ DE BORDEAUX, TALENCE, FRANCE) Recyclable Catalytic Dendrimer Nanoreactor for Part-Per-Million Cu^I Catalysis of 'Click' Chemistry in Water *J. Am. Chem. Soc.* **2014**, *136*, 12092–12098.

Dendritic Copper Catalysts for Homogeneous Click Chemistry in Water



Significance: The dendrimer **1** consists of water/ organic compatible 27 triethyleneglycol (TEG) termini and a hydrophobic core (for the preparation of **1**, see: *Chem. Commun.* **2013**, *49*, 8169). [Hexabenzyltren-Cu]Br **3** and **1** promote the click reaction of benzyl azides and phenyl acetylenes in water to give the corresponding triazoles in 89–96% yield. The catalyst **1-Cu**^I (4–200 ppm Cu^I), prepared from CuSO $_4$ ·5H $_2$ O and NaAsc, also catalyzed the click reaction in 81–99% yield.

Comment: The TEG termini render the dendrimer **1** water-soluble, and the hydrophobic core allows solubilization of the hydrophobic compound **3** and the substrates in water. The interaction between **1** and **3** was shown by selective ¹H NMR shifts and a NOESY spectrum. The micelle nanoreactor **1** was recycled ten times (10th reuse: **2a**, 91% yield). The click reaction was performed with 1 ppm **1-Cu¹** to give **2a** in 50% yield, whose turnover number (TON) and turnover frequency (TOF) were 510000 and 21200 h⁻¹, respectively.

SYNFACTS Contributors: Yasuhiro Uozumi, Yoichi M. A. Yamada, Rikako Ishii Synfacts 2014, 10(12), 1331 Published online: 18.11.2014 **DOI:** 10.1055/s-0034-1379574; **Reg-No.:** Y13214SF