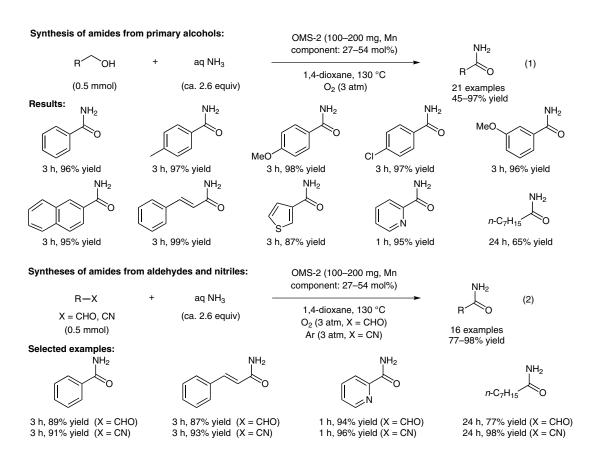
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Heterogeneously Catalyzed Synthesis of Primary Amides Directly from Primary Alcohols and Aqueous Ammonia *Angew. Chem. Int. Ed.* **2012**, *51*, 544–547.

## **Synthesis of Primary Amines with OMS-2**



**Significance:** Manganese oxide based octahedral molecular sieves (OMS-2) catalyzed the reaction of primary alcohols with aqueous ammonia to give the corresponding amides in 65–99% yield under molecular oxygen (10 examples, eq. 1). The reactions of aldehydes and nitriles with aqueous ammonia also proceeded in the presence of OMS-2 to give the corresponding amides in 77–98% yield (16 examples, eq. 2). In the formation of 2-pyridinecarboxamide from 2-pyridinemethanol, the catalyst was recovered by filtration and reused eleven times without significant loss of its catalytic activity (1<sup>st</sup> reuse: 93% yield, 11<sup>th</sup> reuse: 85% yield).

**Comment:** Suib and co-workers have previously reported the preparation of OMS-2 (*Chem. Mater.* **1994**, *6*, 815). In the formation of benzamide from benzyl alcohol, the catalytic activity of OMS-2 was superior to that of precursors of OMS-2 (KMnO<sub>4</sub>, MnSO<sub>4</sub>·H<sub>2</sub>O), other manganese-based oxides (MnO<sub>2</sub> and Mn<sub>3</sub>O<sub>4</sub>) and other metal oxides (Co<sub>3</sub>O<sub>4</sub>, CeO<sub>2</sub>). After the reaction of benzyl alcohol with aqueous ammonia, no leaching of manganese species was observed by ICP-AES analysis.

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Category

Polymer-Supported Synthesis

**Key words** 

amides

manganese oxide

heterogeneous catalysis

