

Supporting Information  
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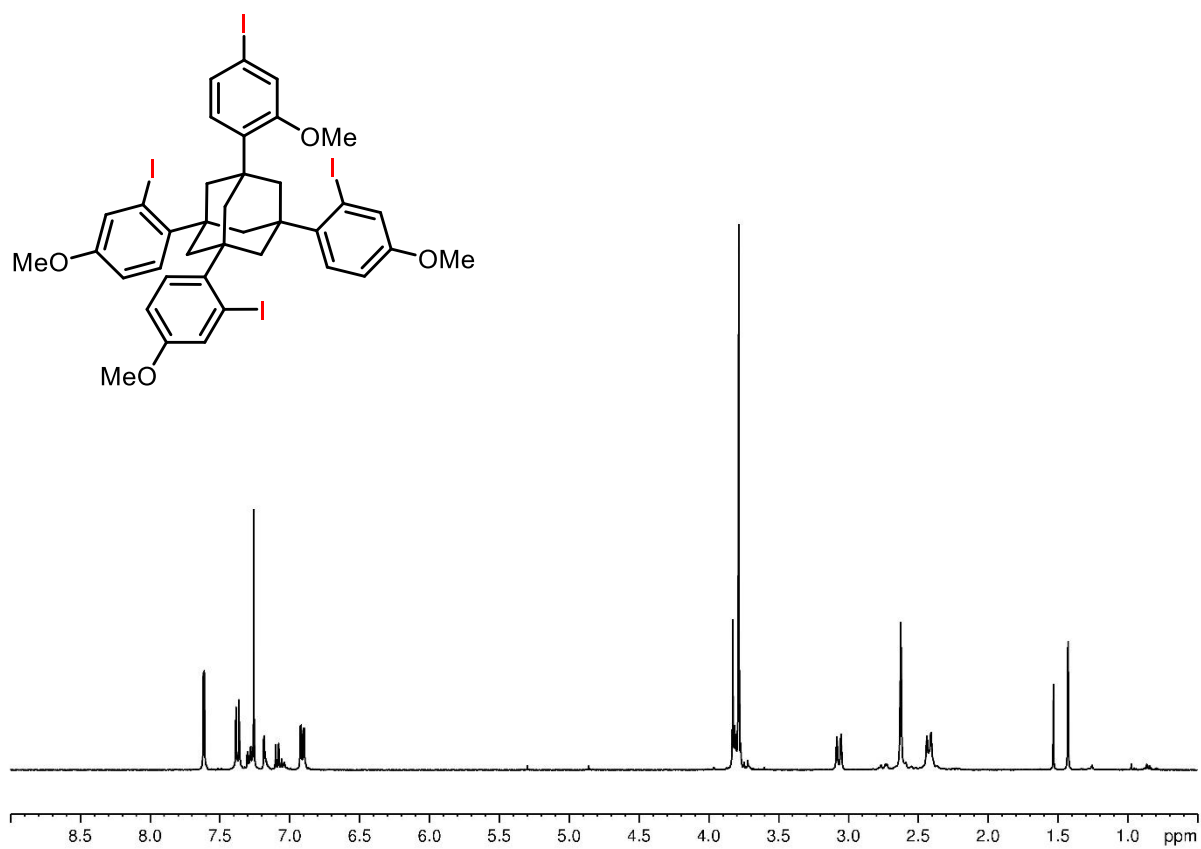
# Supporting Information

for manuscript entitled

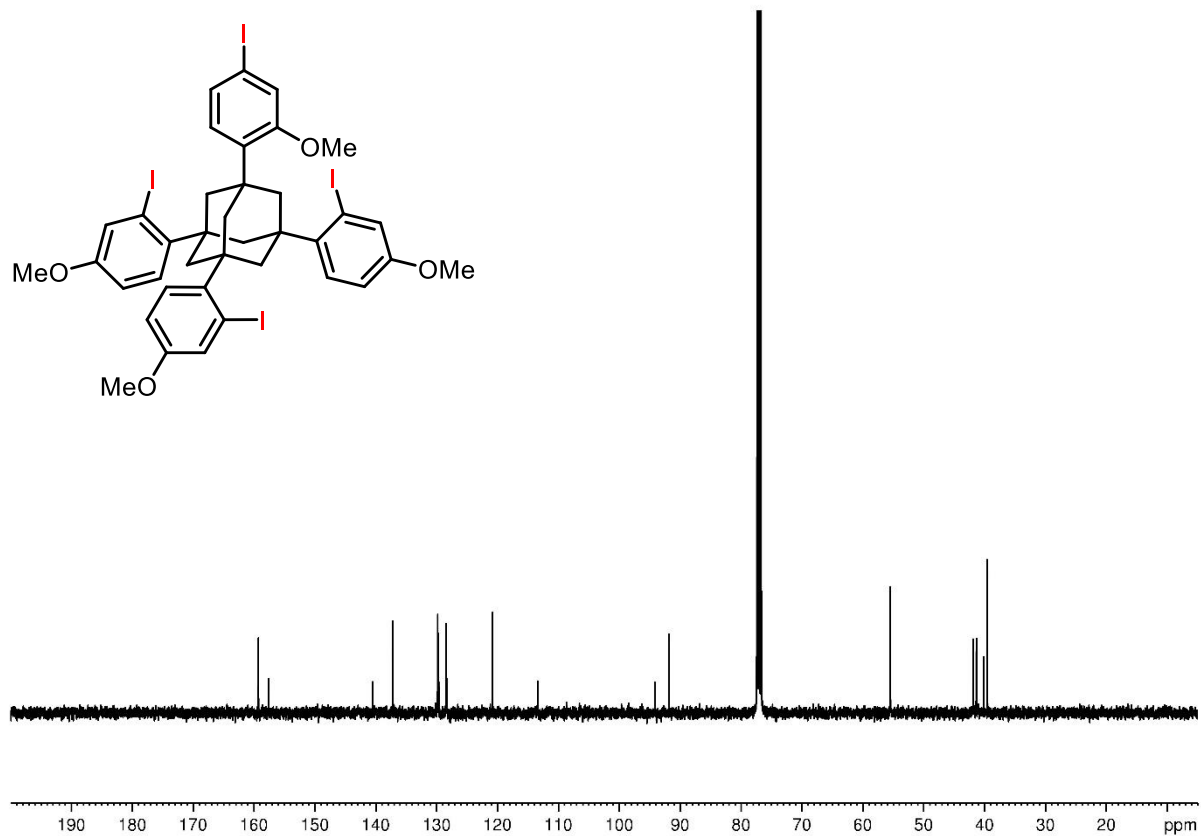
Kinetic or Thermodynamic Product? Case Studies on the Formation of Regioisomers of  
Tetraphenyladamantanes

by Tim Berking, Wolfgang Frey and Clemens Richert

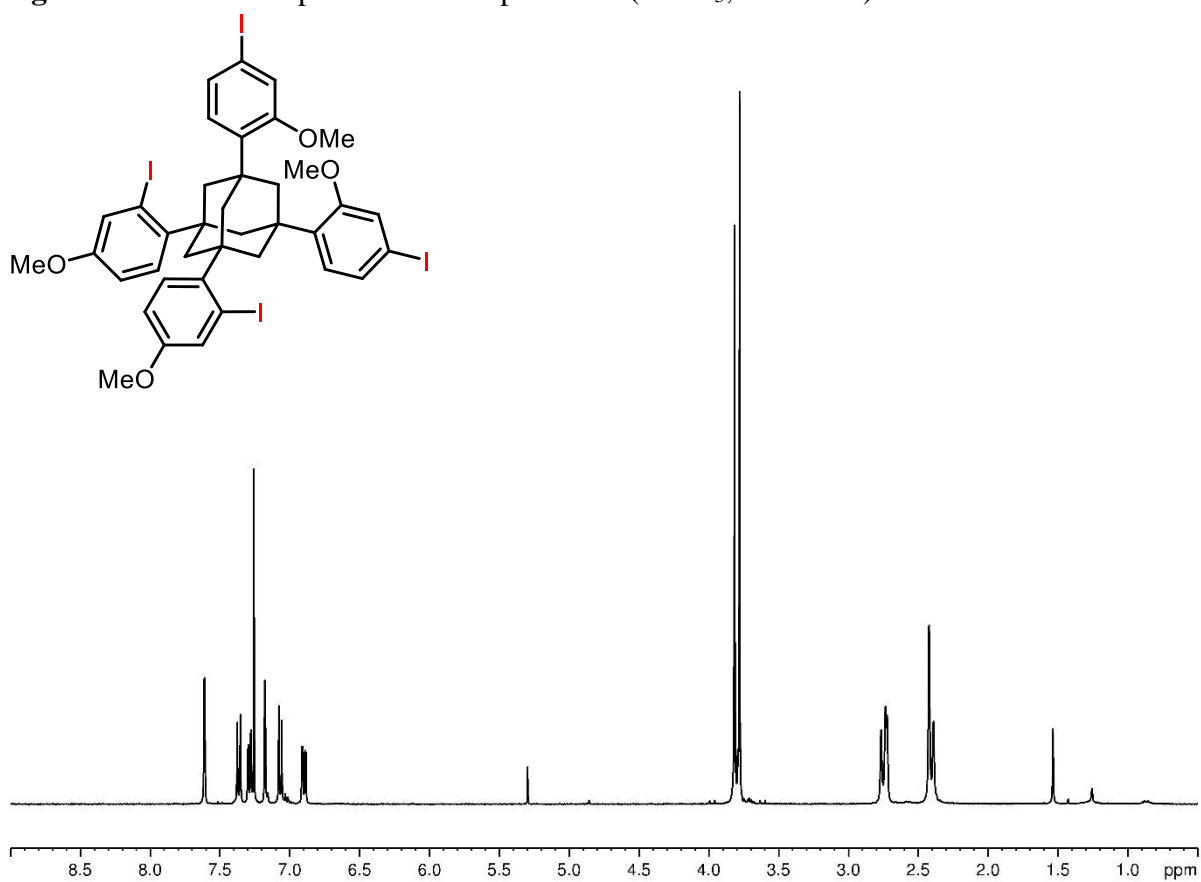
## 1. NMR Spectra



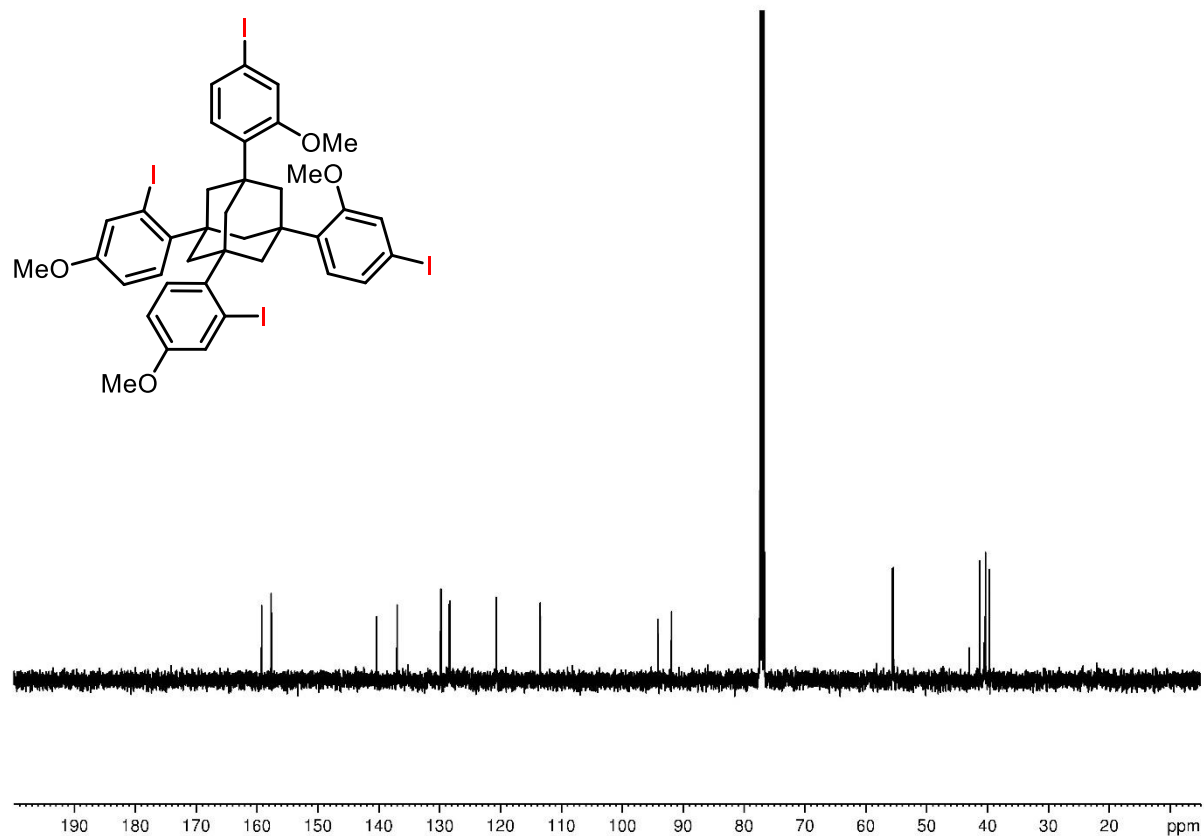
**Figure S1.**  $^1\text{H-NMR}$  spectrum of compound **11** ( $\text{CDCl}_3$ , 400 MHz).



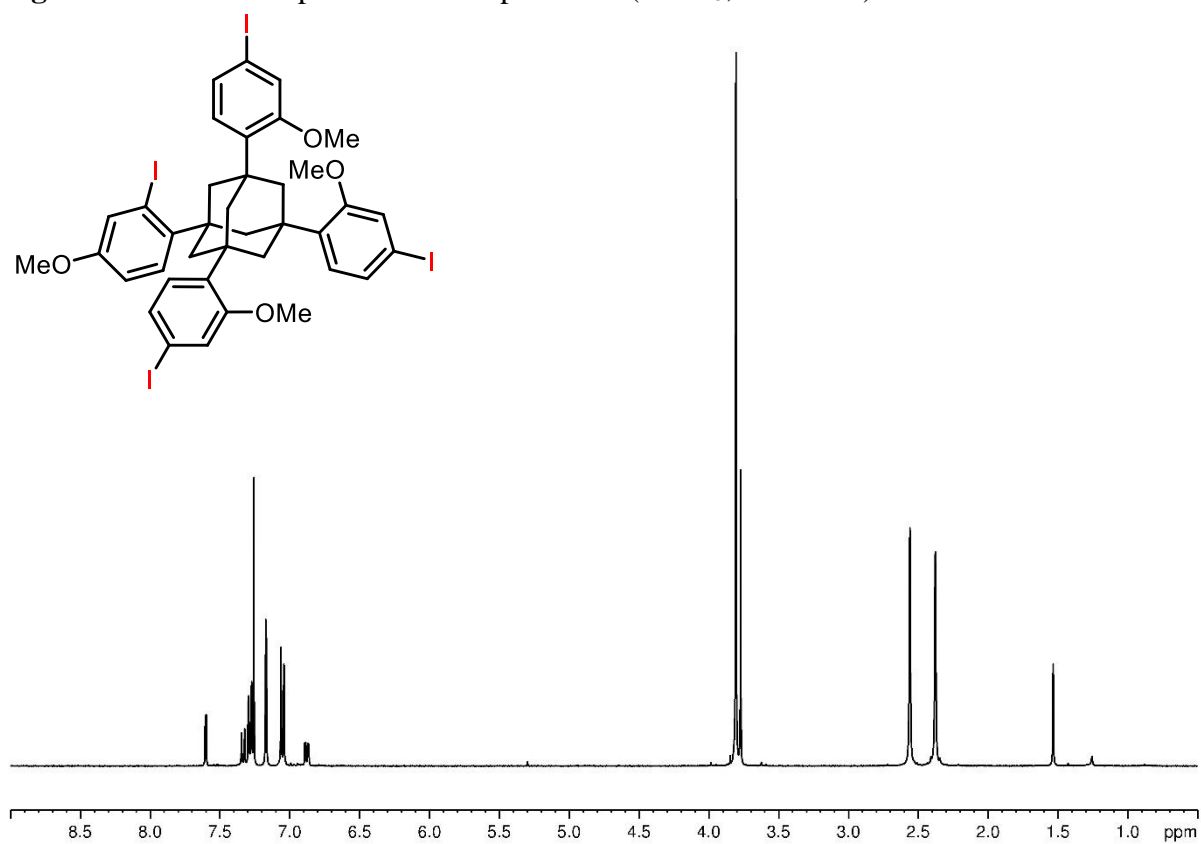
**Figure S2.**  $^{13}\text{C}$ -NMR spectrum of compound **11** ( $\text{CDCl}_3$ , 101 MHz).



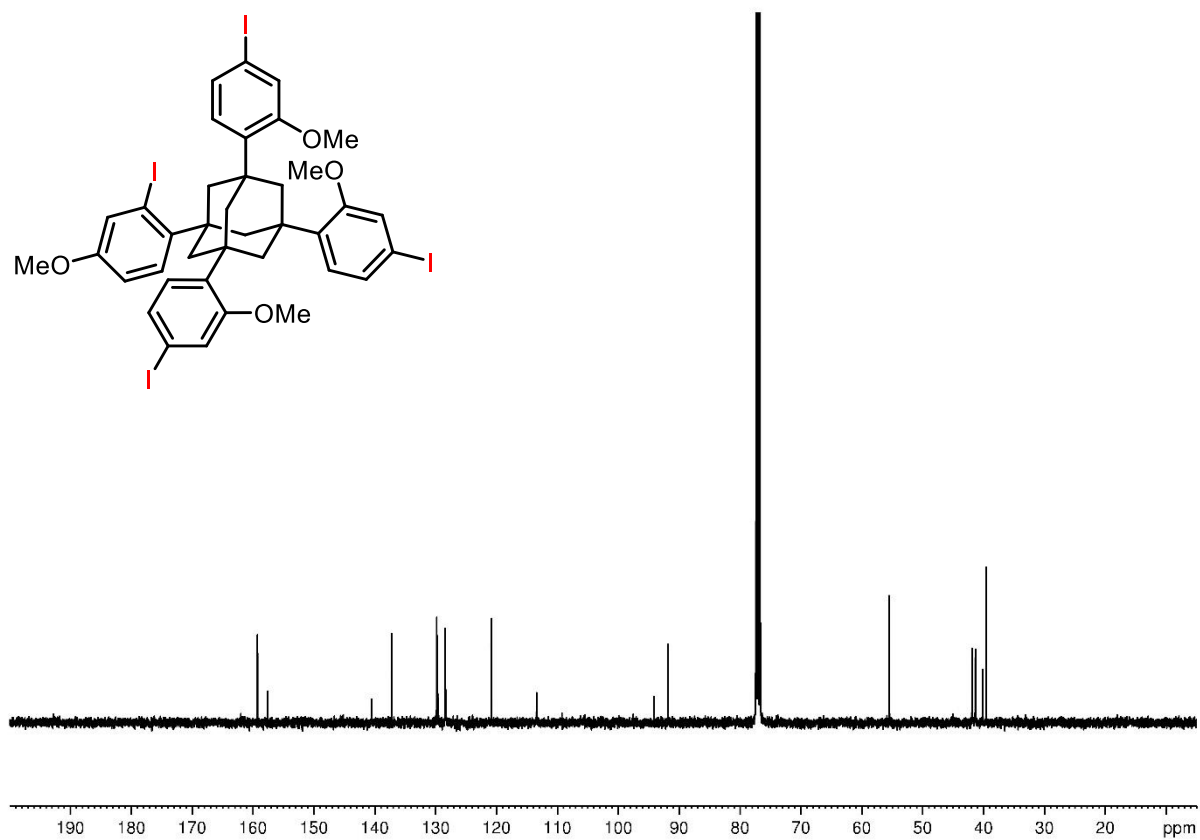
**Figure S3.**  $^1\text{H}$ -NMR spectrum of compound **12** ( $\text{CDCl}_3$ , 400 MHz).



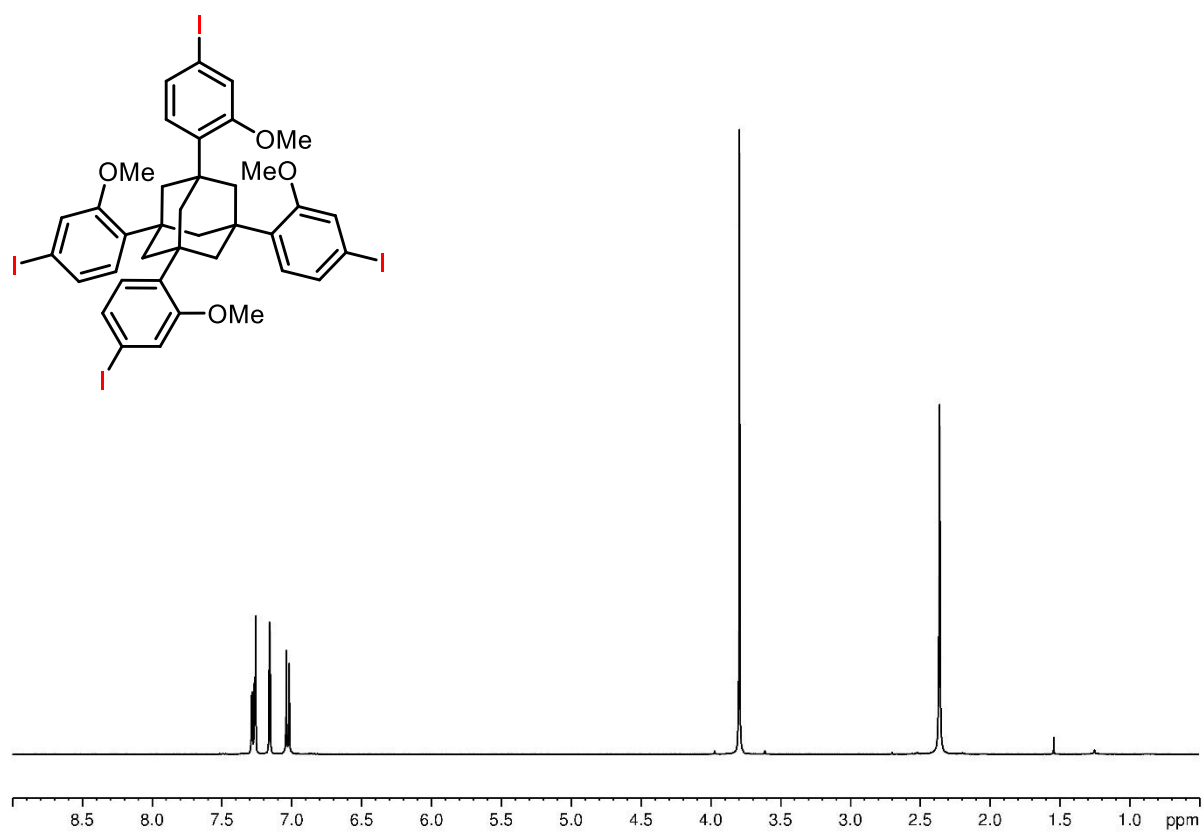
**Figure S4.** <sup>13</sup>C-NMR spectrum of compound **12** (CDCl<sub>3</sub>, 101 MHz).



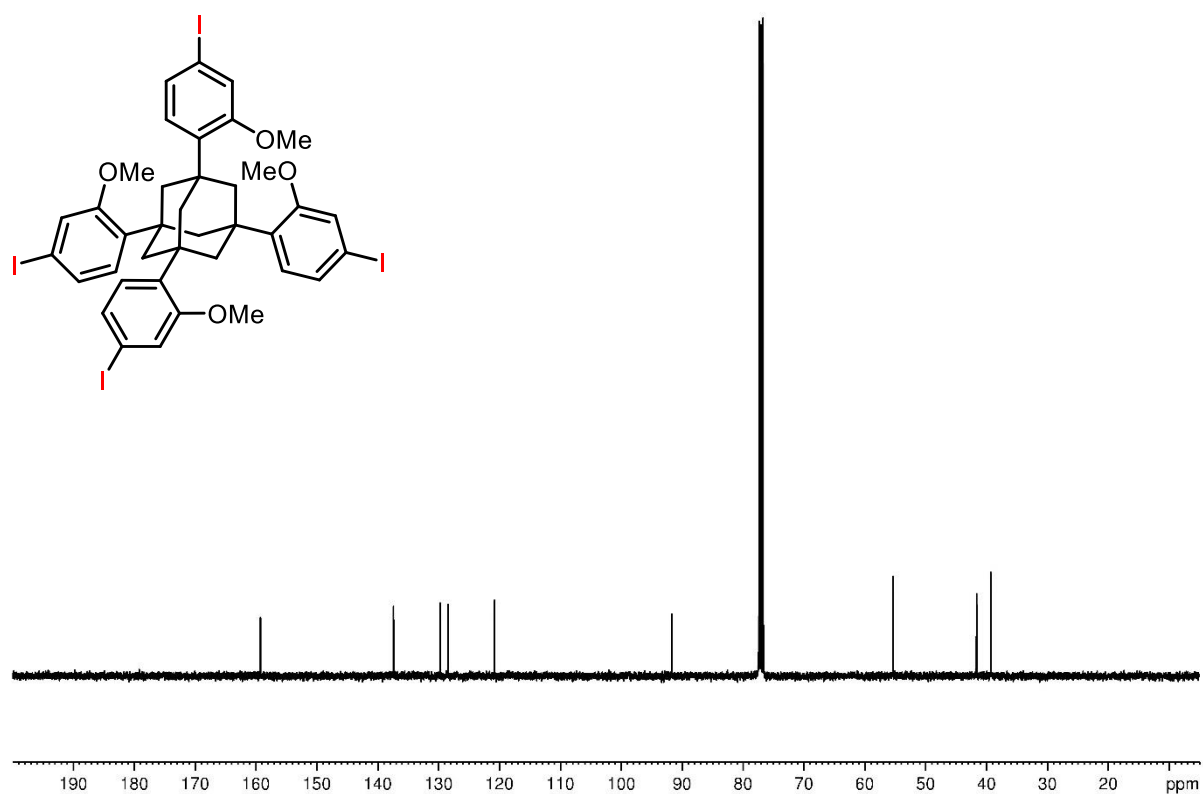
**Figure S5.** <sup>1</sup>H-NMR spectrum of compound **13** (CDCl<sub>3</sub>, 400 MHz).



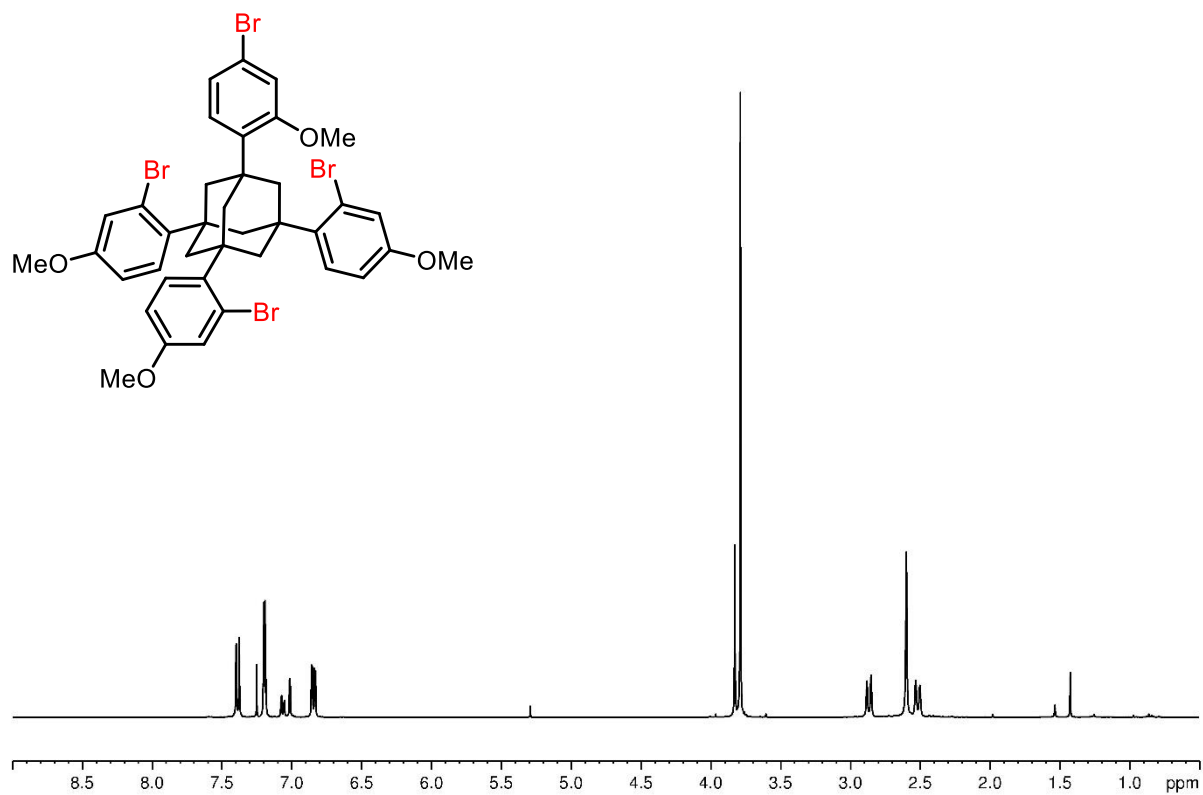
**Figure S6.**  $^{13}\text{C}$ -NMR spectrum of compound **13** ( $\text{CDCl}_3$ , 101 MHz).



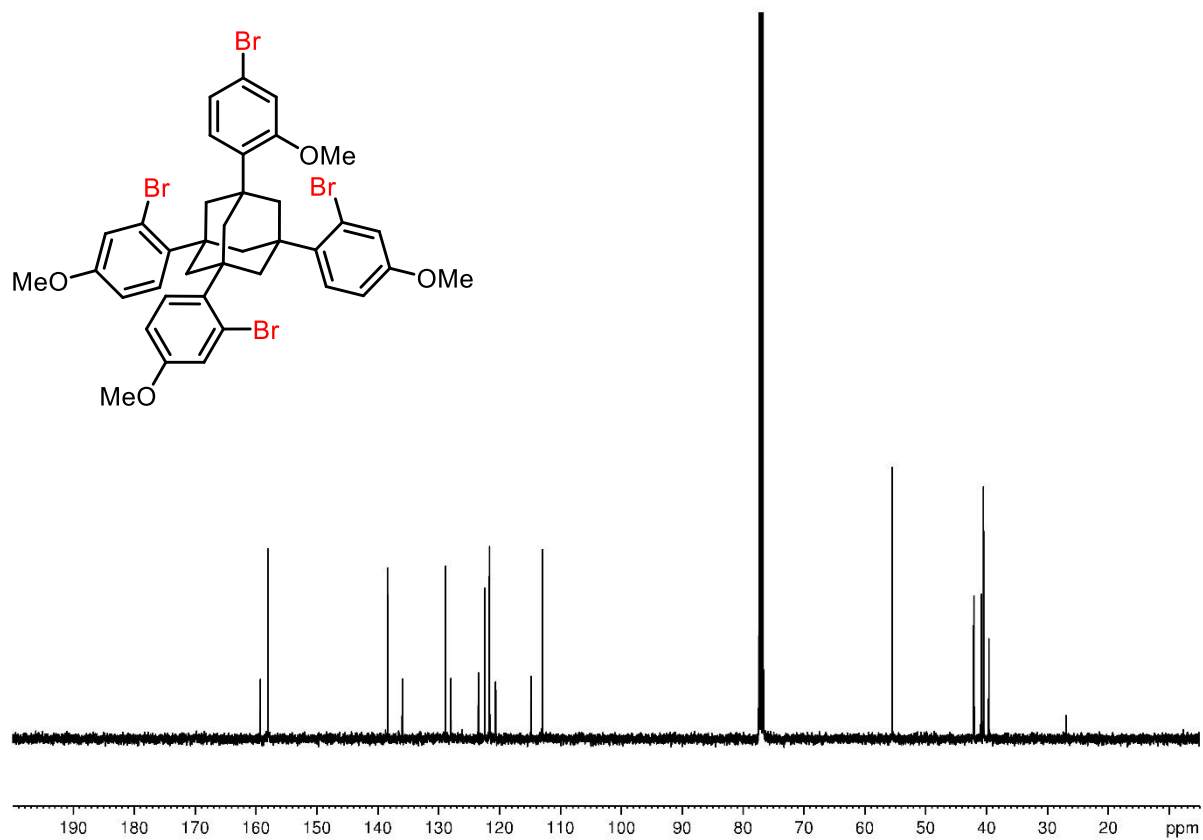
**Figure S7.**  $^1\text{H}$ -NMR spectrum of iTIM ( $\text{CDCl}_3$ , 400 MHz).



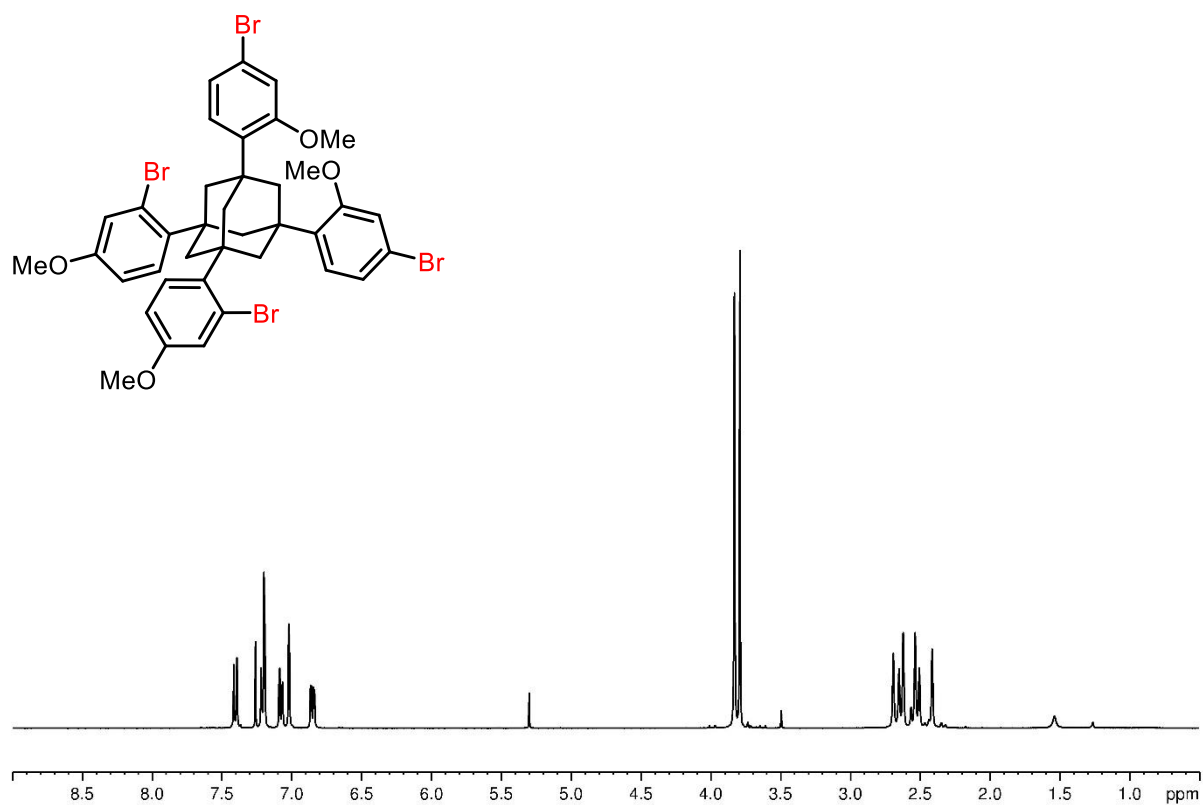
**Figure S8.** <sup>13</sup>C-NMR spectrum of iTIM (CDCl<sub>3</sub>, 101 MHz).



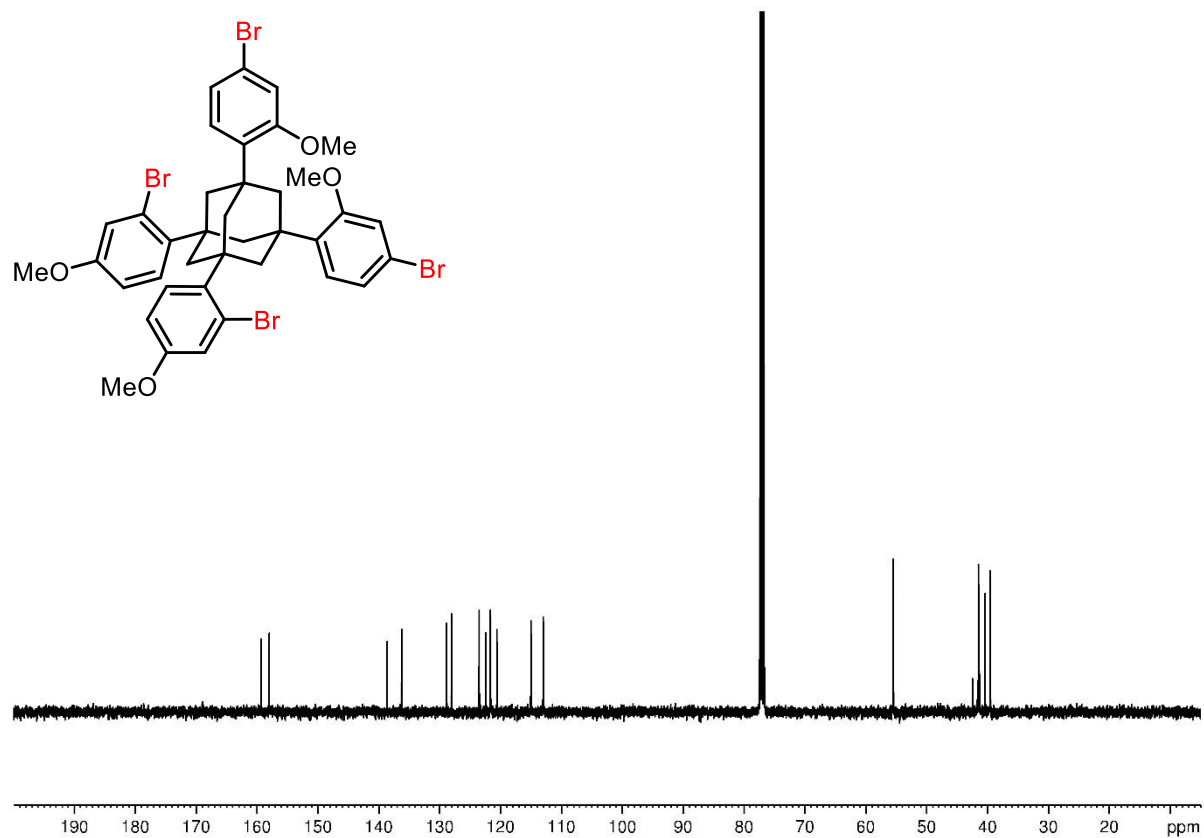
**Figure S9.** <sup>1</sup>H-NMR spectrum of compound 5 (CDCl<sub>3</sub>, 400 MHz).



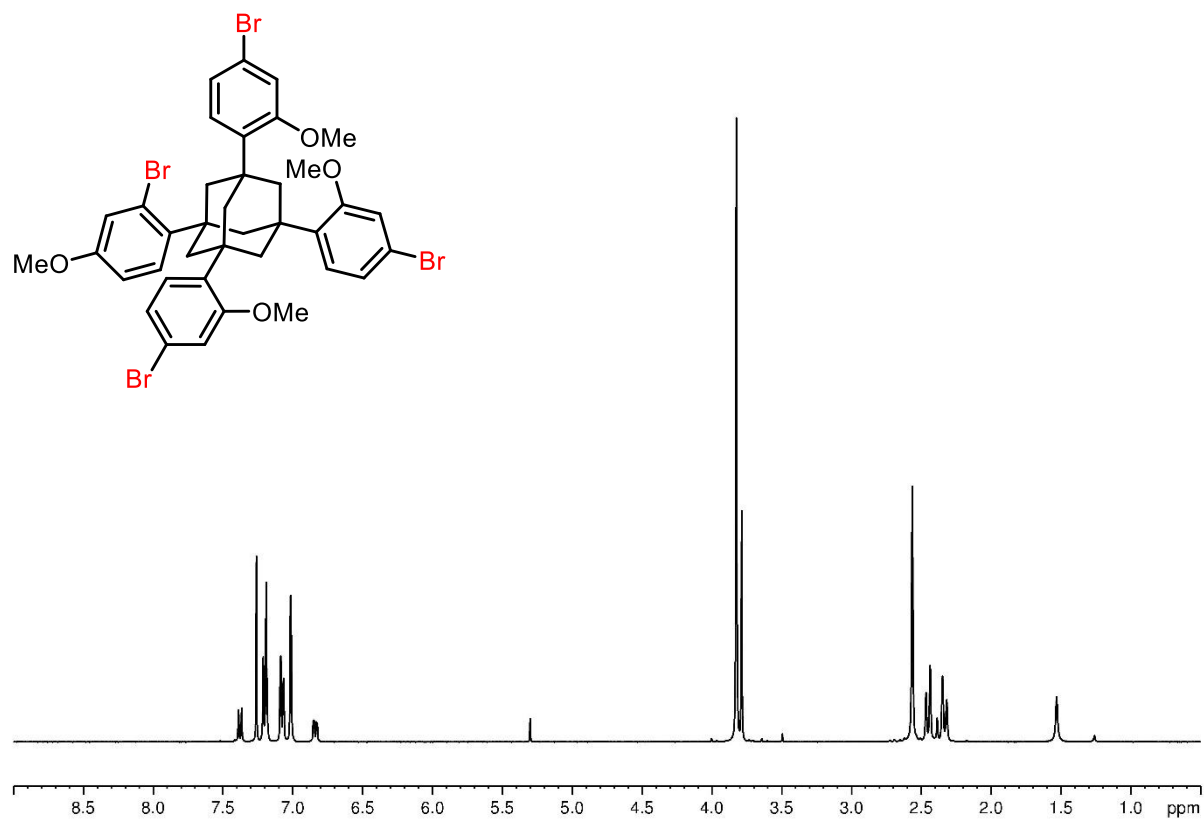
**Figure S10.** <sup>13</sup>C-NMR spectrum of compound **5** (CDCl<sub>3</sub>, 101 MHz).



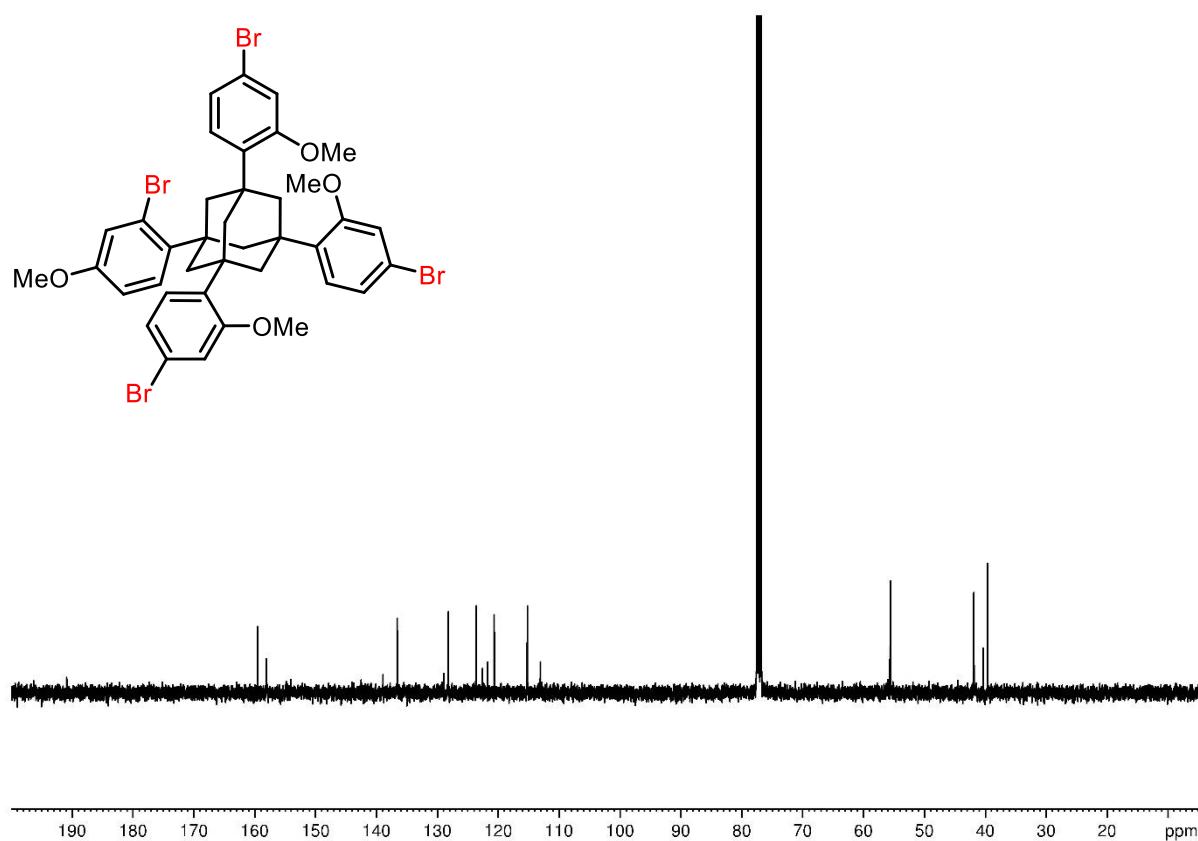
**Figure S11.** <sup>1</sup>H-NMR spectrum of compound **6** (CDCl<sub>3</sub>, 400 MHz).



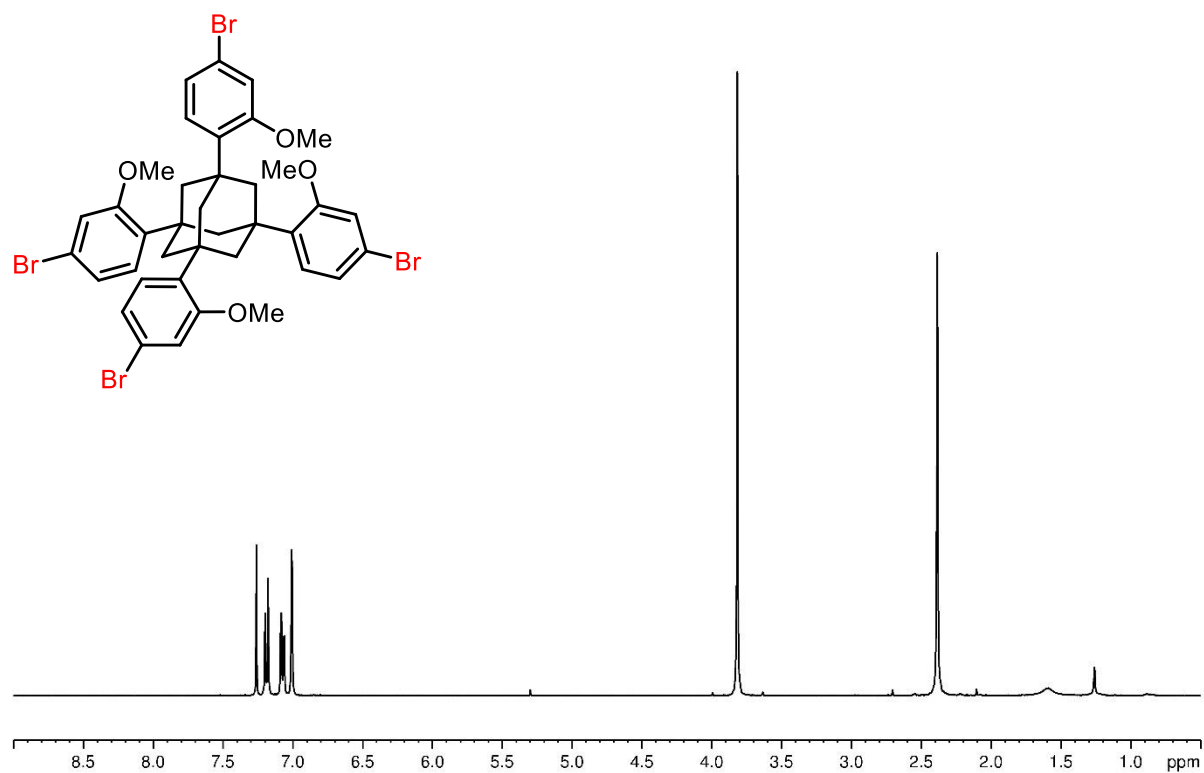
**Figure S12.** <sup>13</sup>C-NMR spectrum of compound 6 (CDCl<sub>3</sub>, 101 MHz).



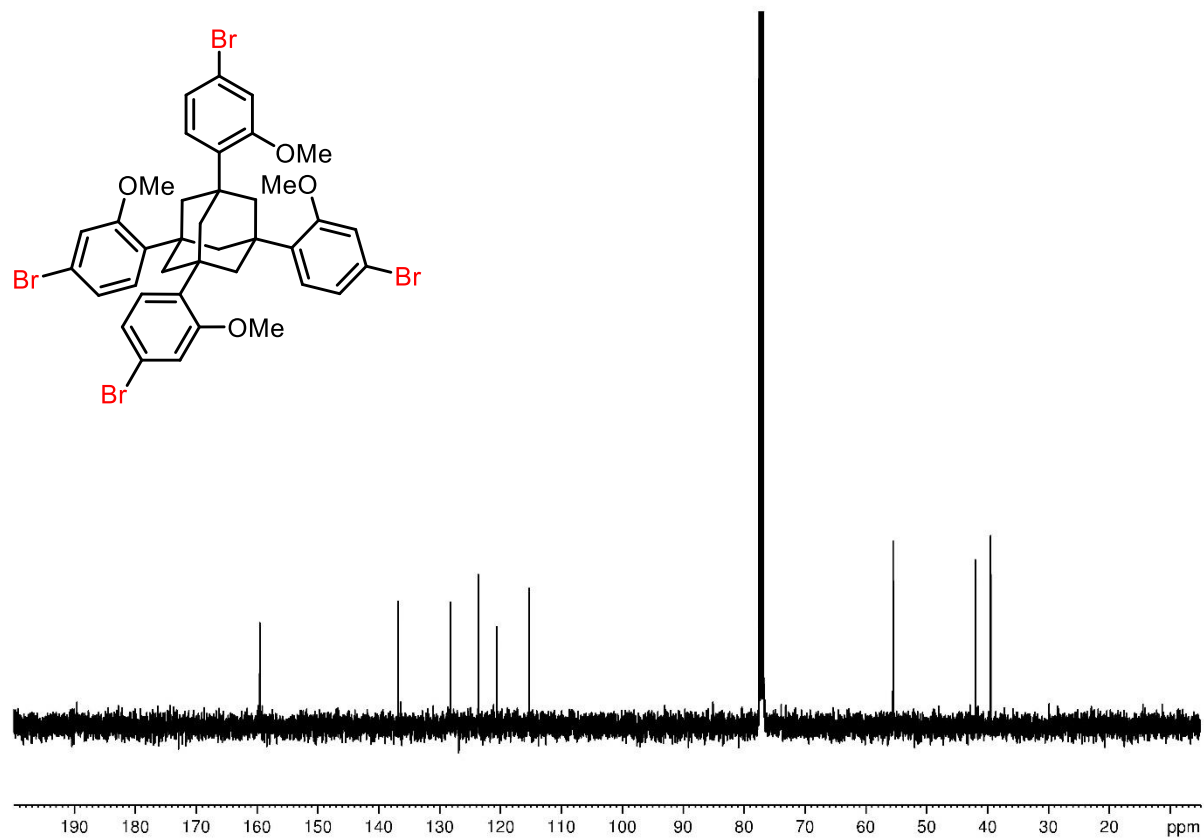
**Figure S13.** <sup>1</sup>H-NMR spectrum of compound 7 (CDCl<sub>3</sub>, 400 MHz).



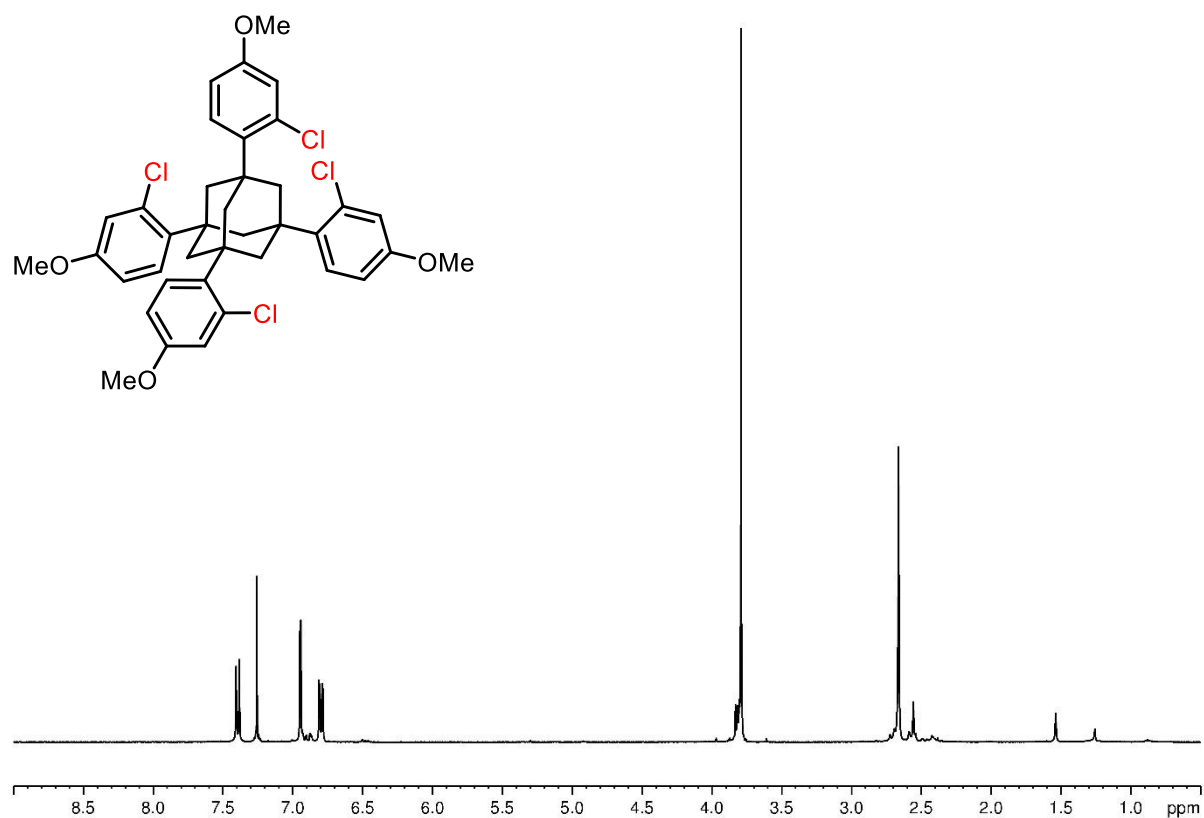
**Figure S14.**  $^{13}\text{C}$ -NMR spectrum of compound **7** ( $\text{CDCl}_3$ , 101 MHz).



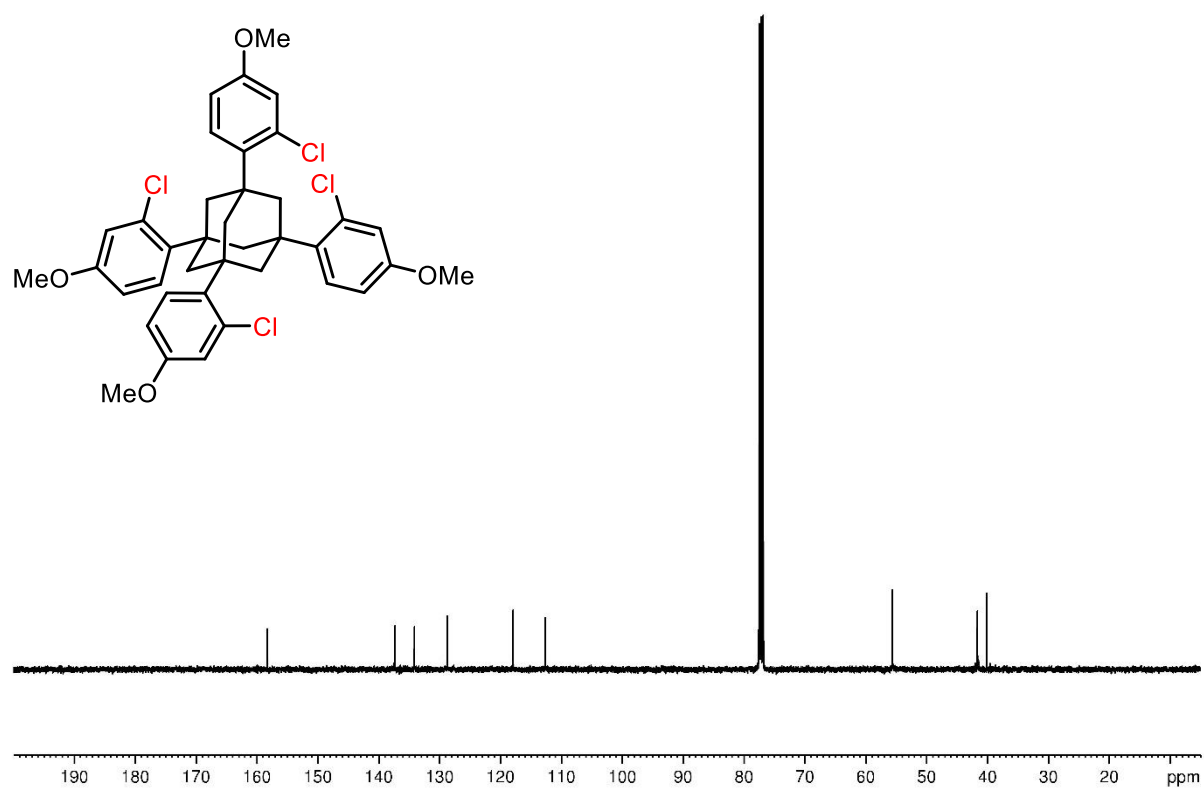
**Figure S15.** <sup>1</sup>H-NMR spectrum of iTBro (CDCl<sub>3</sub>, 400 MHz).



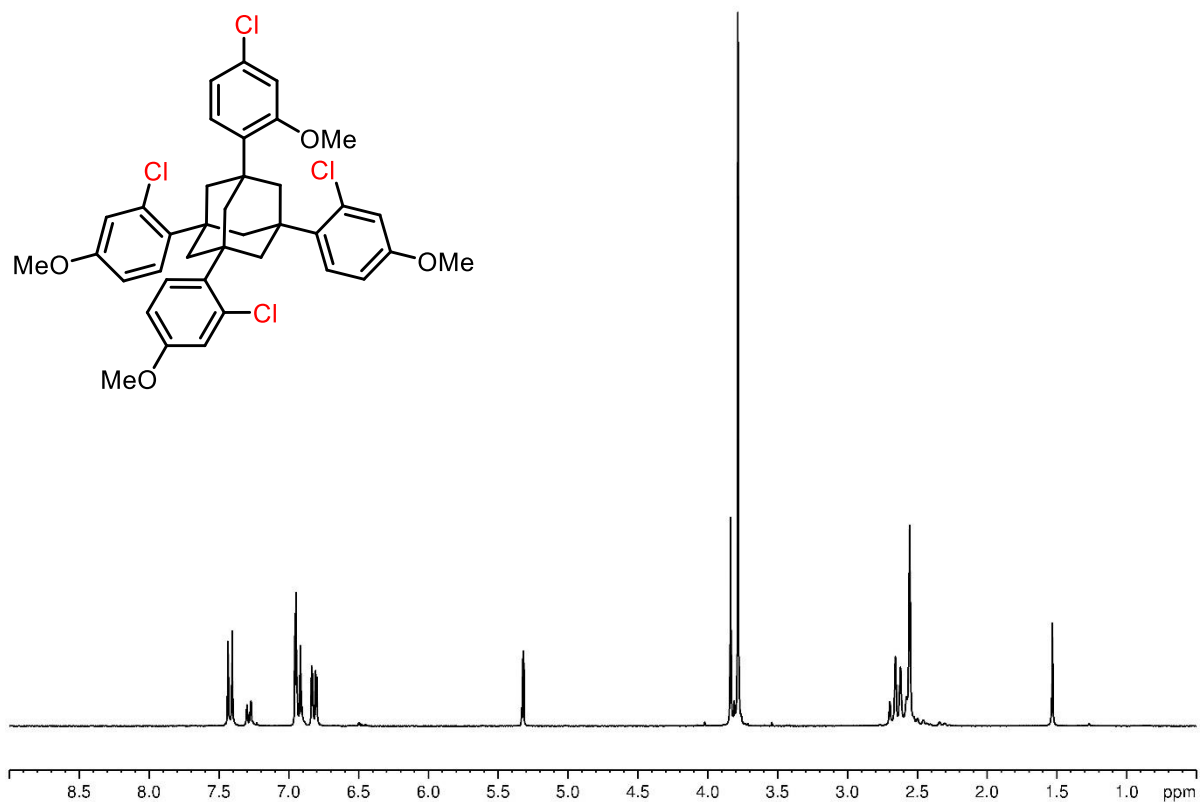
**Figure S16.** <sup>13</sup>C-NMR spectrum of iTBro (CDCl<sub>3</sub>, 101 MHz).



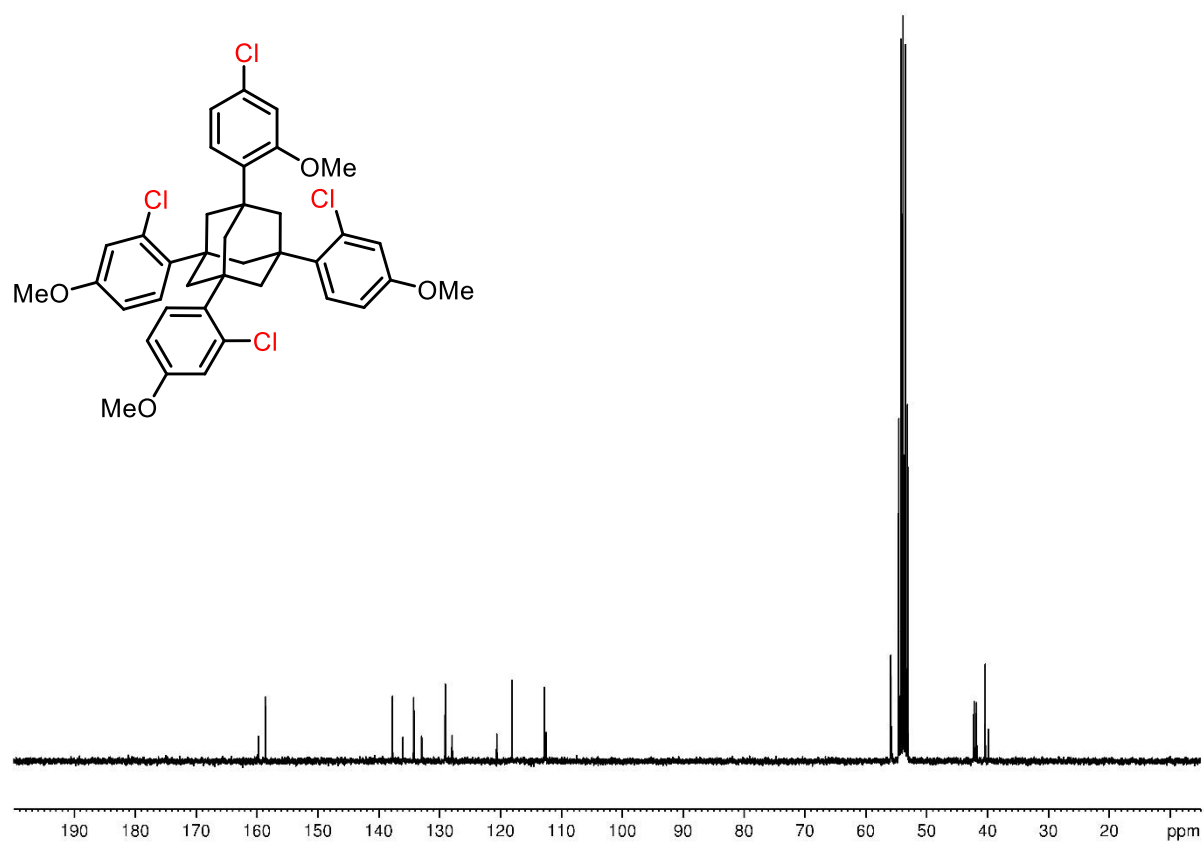
**Figure S17.** <sup>1</sup>H-NMR spectrum of TCIM (CDCl<sub>3</sub>, 400 MHz).



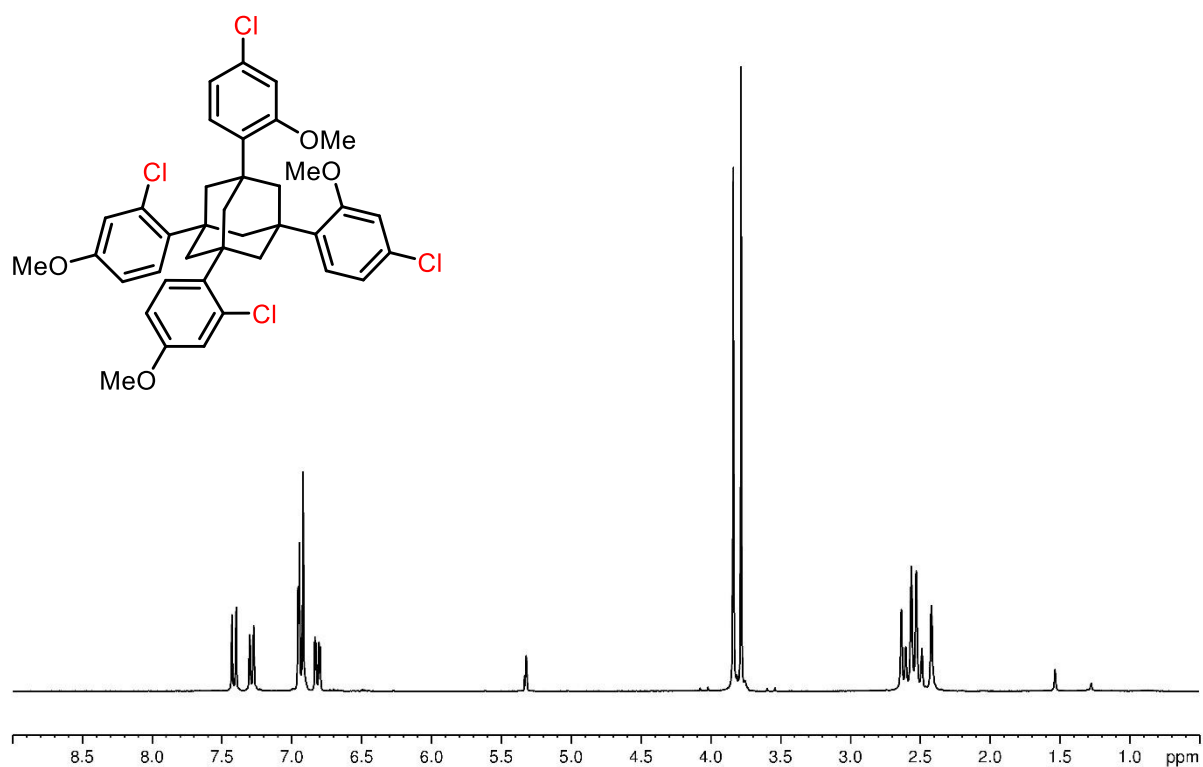
**Figure S18.** <sup>13</sup>C-NMR spectrum of Compound TCIM (CDCl<sub>3</sub>, 101 MHz).



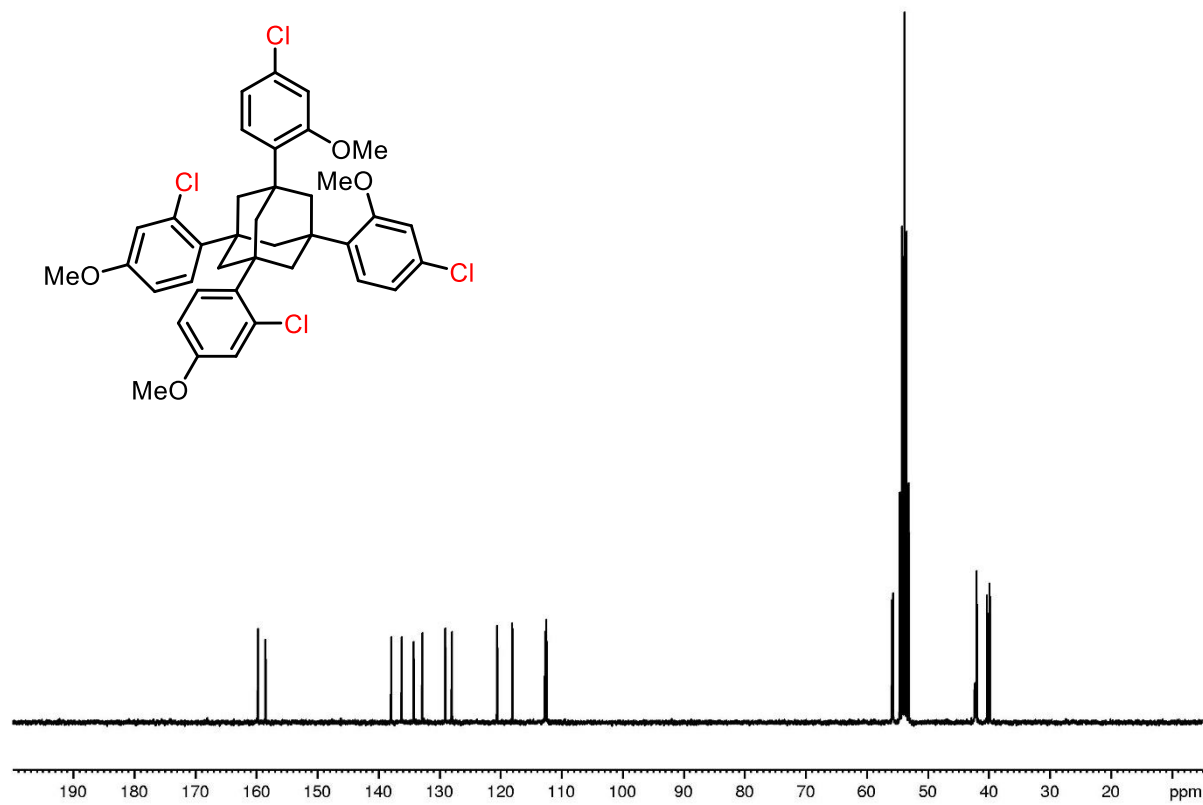
**Figure S19.** <sup>1</sup>H-NMR spectrum of compound **16** (CD<sub>2</sub>Cl<sub>2</sub>, 300 MHz).



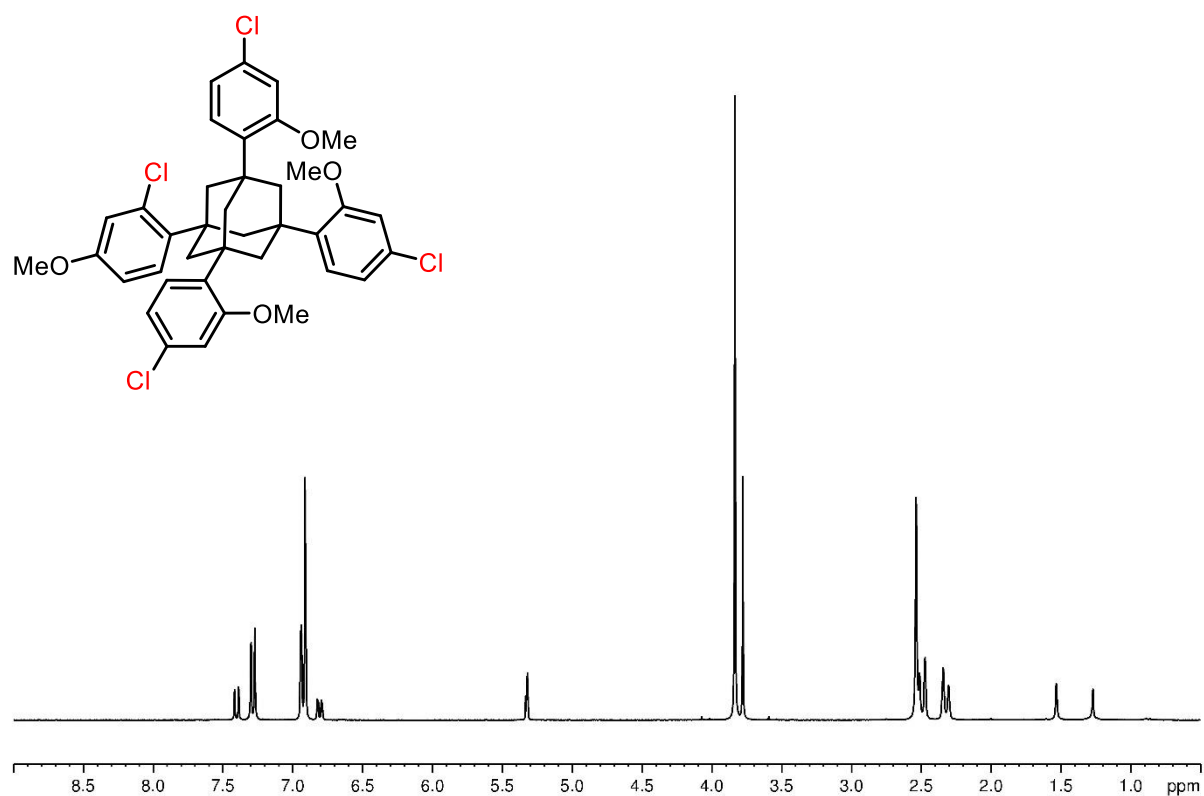
**Figure S20.** <sup>13</sup>C-NMR spectrum of compound **16** (CD<sub>2</sub>Cl<sub>2</sub>, 75 MHz).



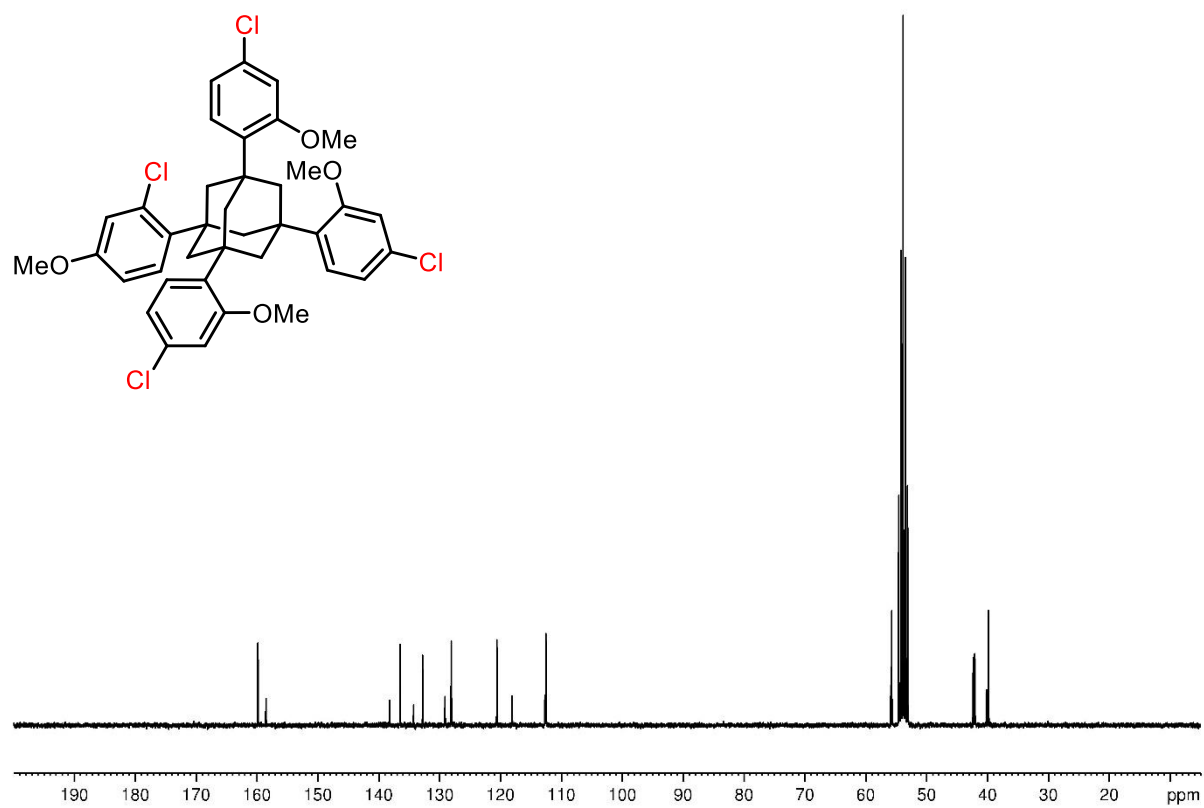
**Figure S21.** <sup>1</sup>H-NMR spectrum of compound **17** (CD<sub>2</sub>Cl<sub>2</sub>, 300 MHz).



**Figure S22.** <sup>13</sup>C-NMR spectrum of compound **17** (CDCl<sub>3</sub>, 75 MHz).



**Figure S23.** <sup>1</sup>H-NMR spectrum of compound **18** (CD<sub>2</sub>Cl<sub>2</sub>, 300 MHz).



**Figure S24.** <sup>13</sup>C-NMR spectrum of compound **18** (CD<sub>2</sub>Cl<sub>2</sub>, 75 MHz).

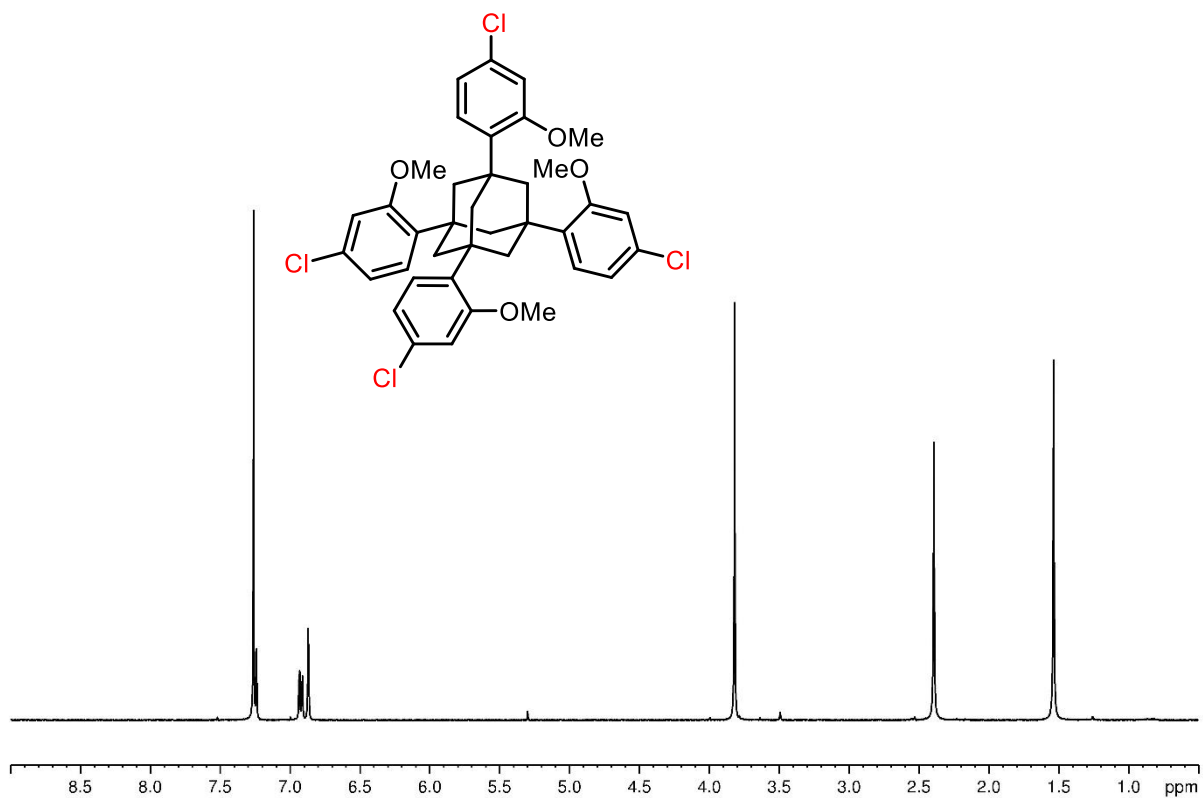


Figure S25. <sup>1</sup>H-NMR spectrum of iTCIM (CDCl<sub>3</sub>, 400 MHz).

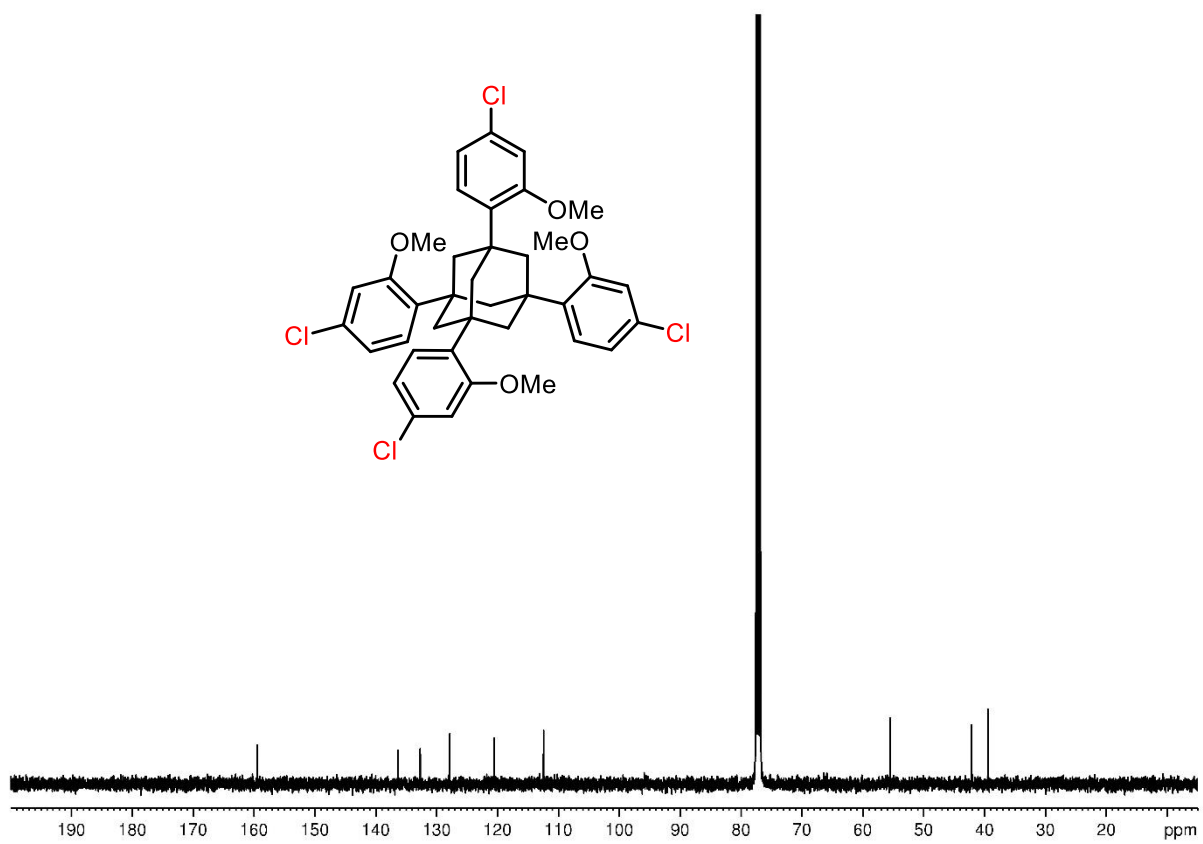
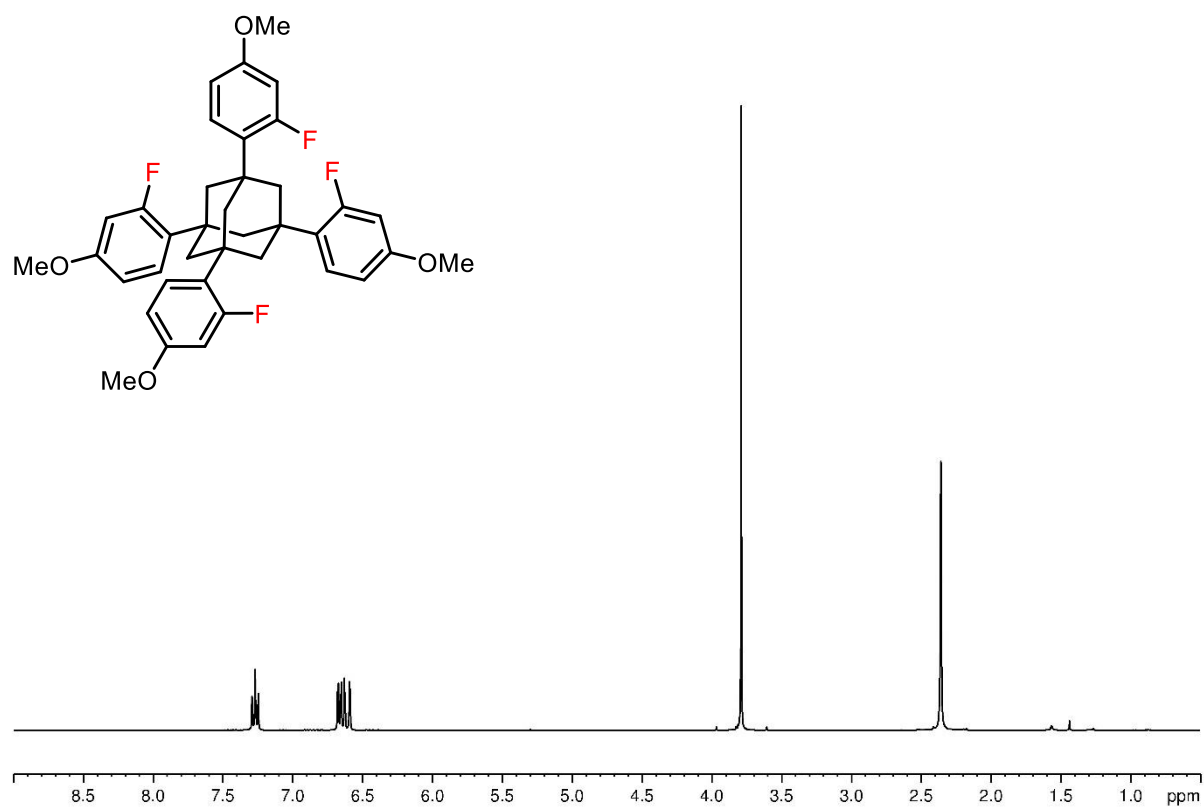
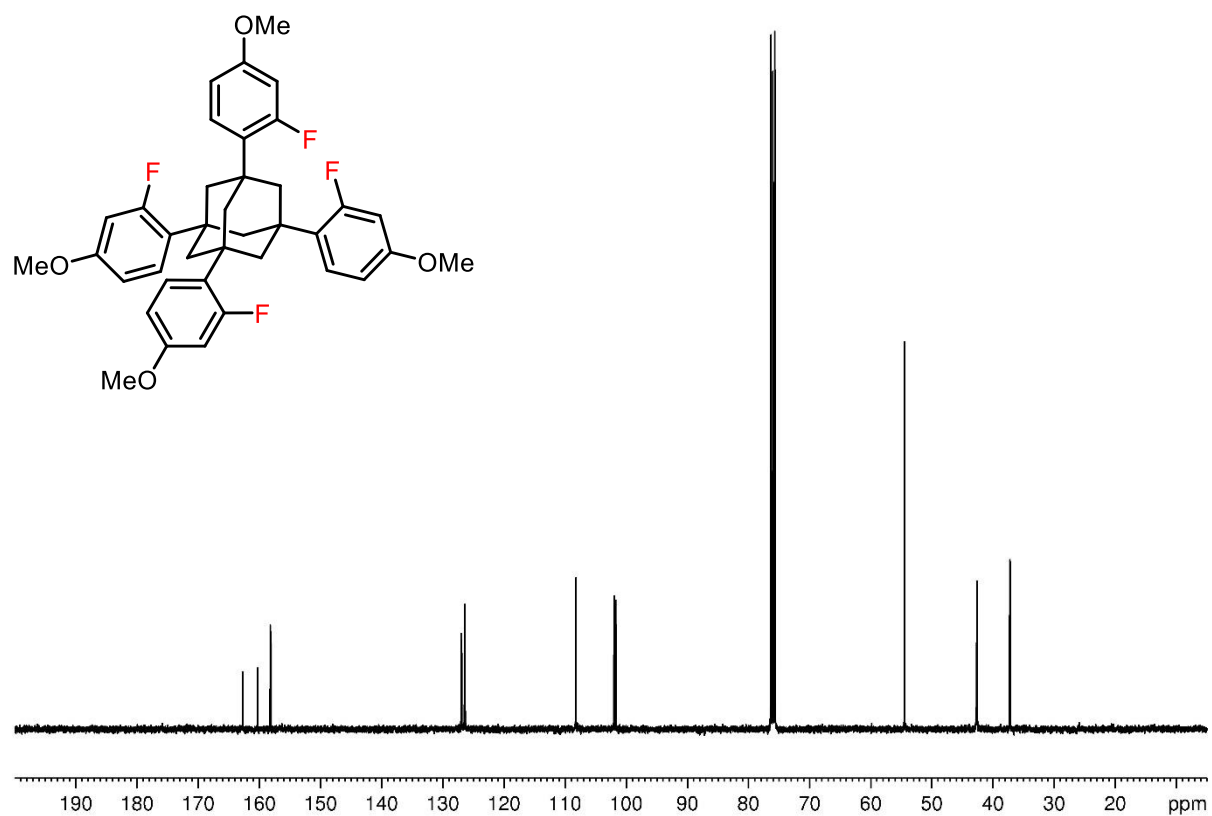


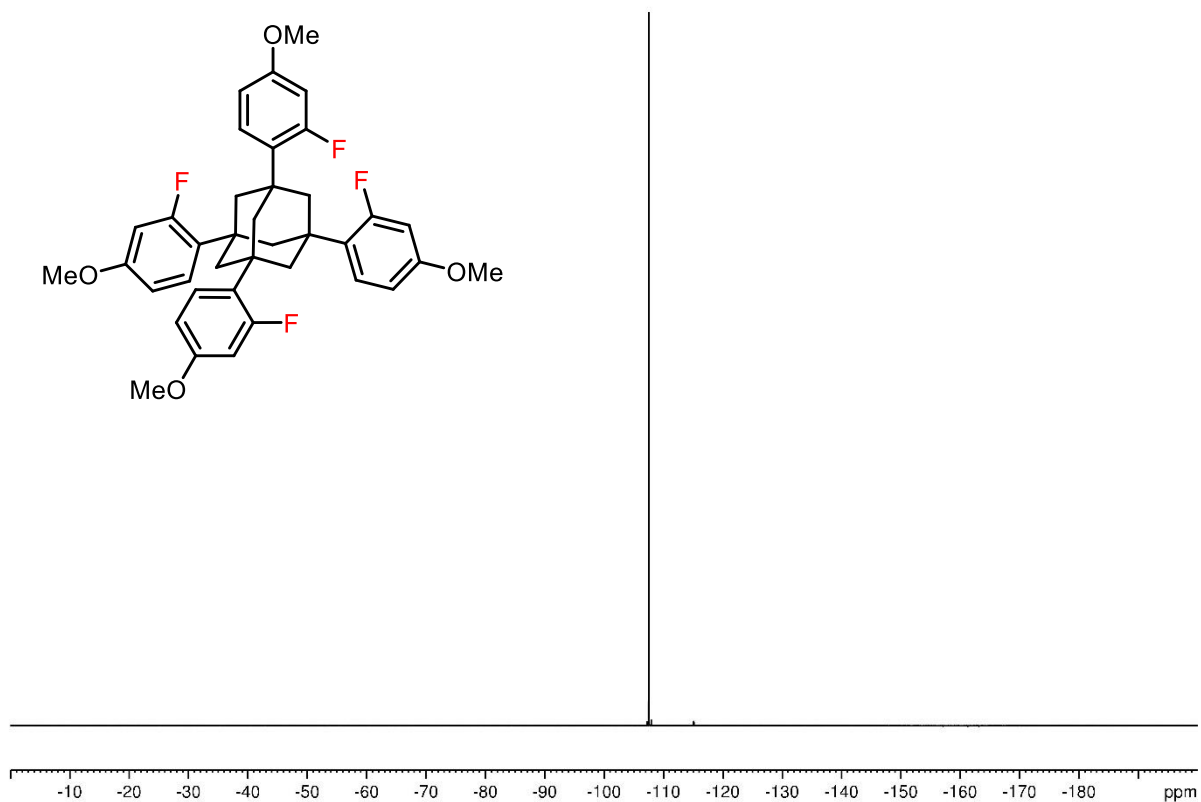
Figure S26. <sup>13</sup>C-NMR spectrum of iTCIM (CDCl<sub>3</sub>, 101 MHz).



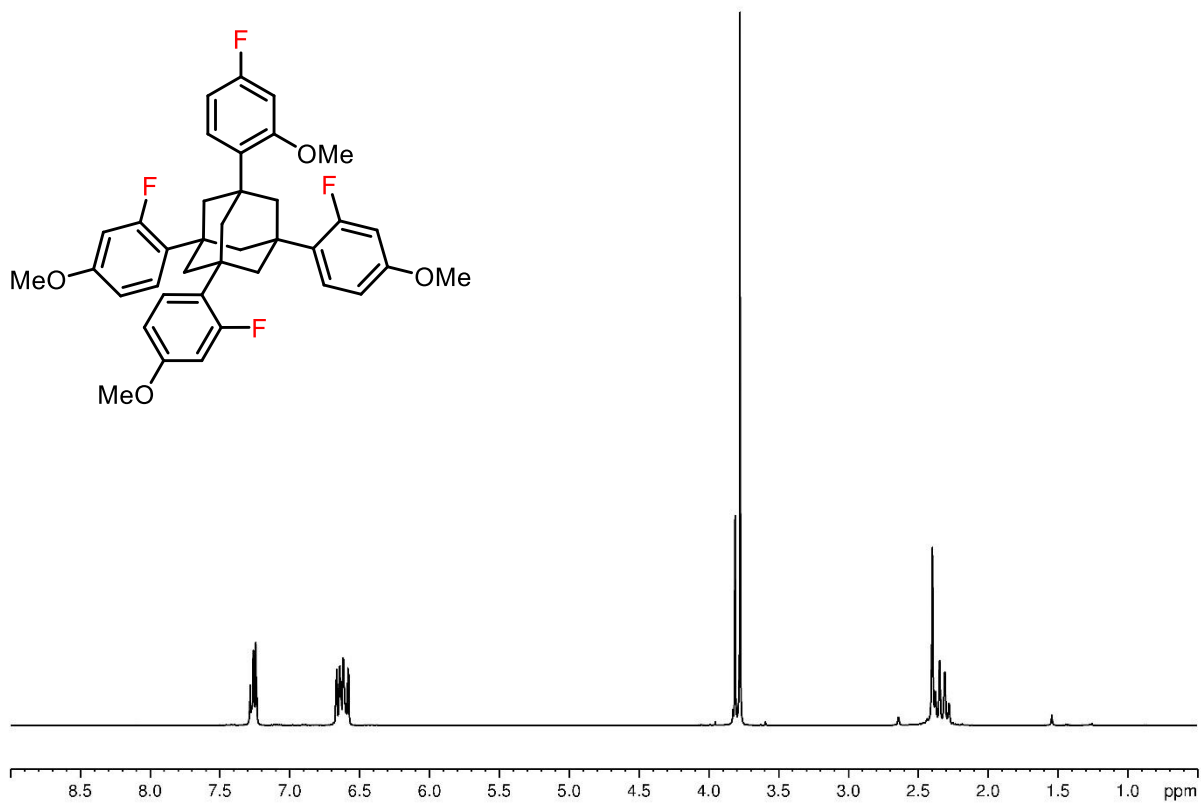
**Figure S27.** <sup>1</sup>H-NMR spectrum of TFM (CDCl<sub>3</sub>, 400 MHz).



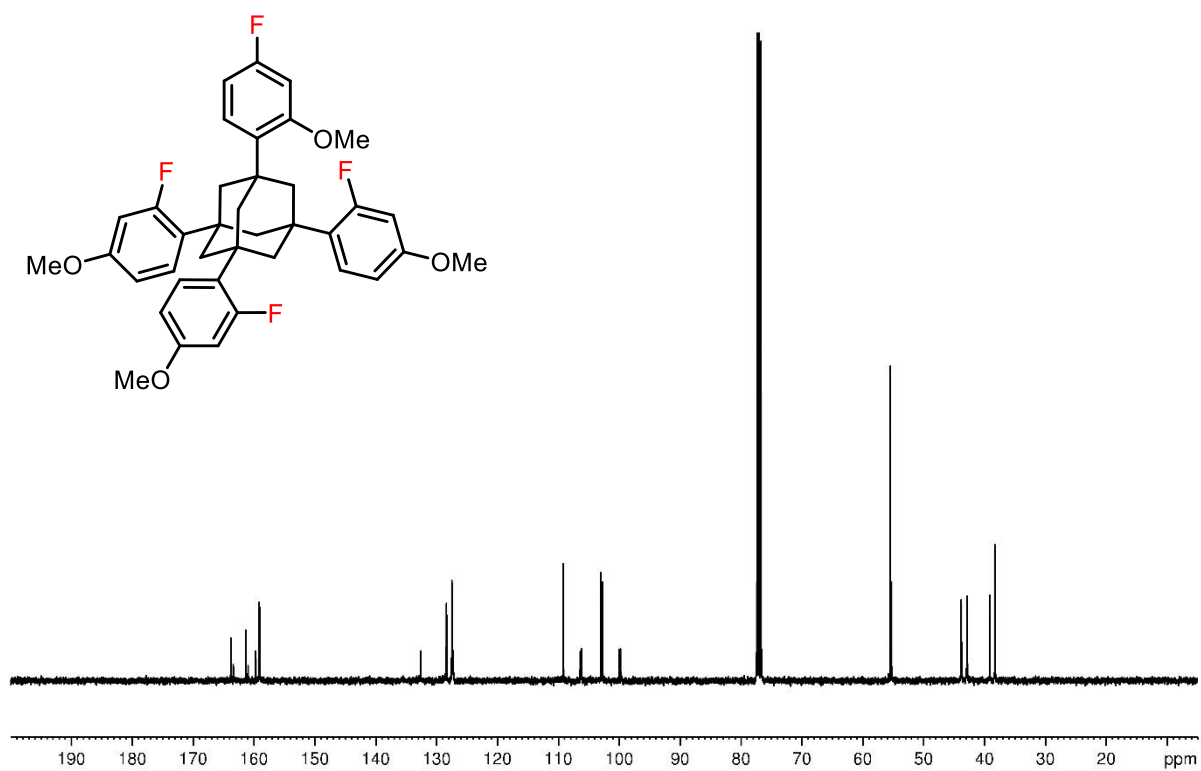
**Figure S28.** <sup>13</sup>C-NMR spectrum of TFM (CDCl<sub>3</sub>, 101 MHz).



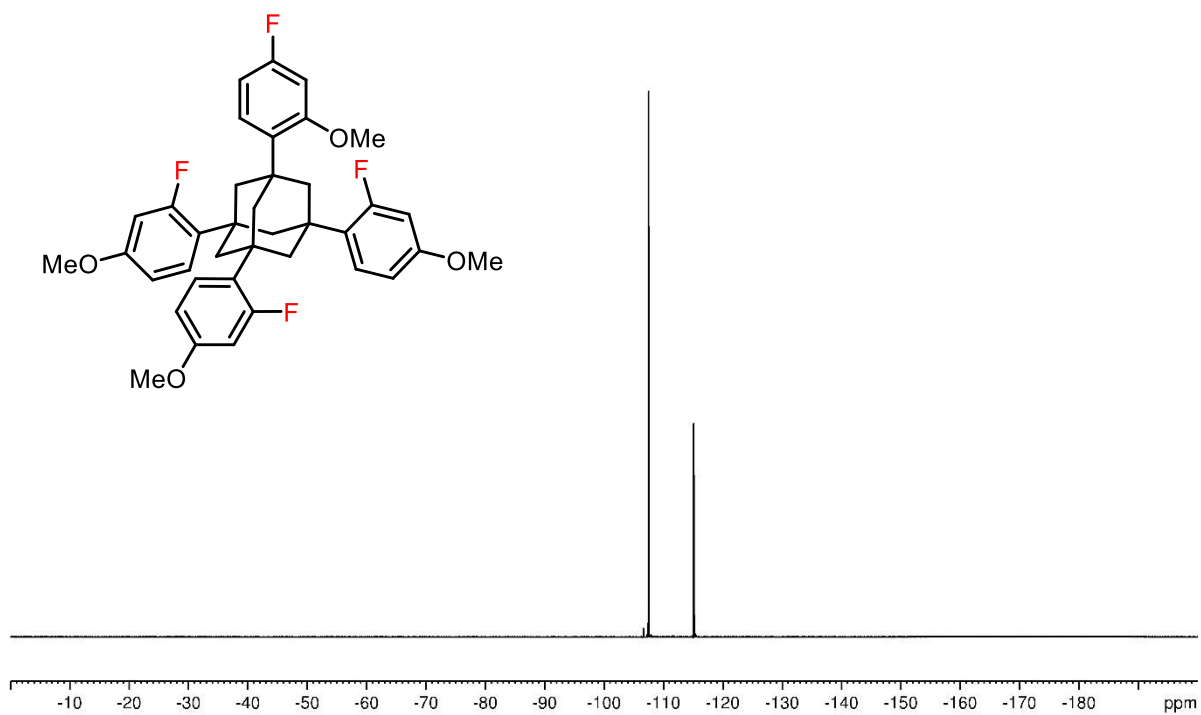
**Figure S29.**  $^{19}\text{F}$ -NMR spectrum of Compound TFM ( $\text{CDCl}_3$ , 376 MHz).



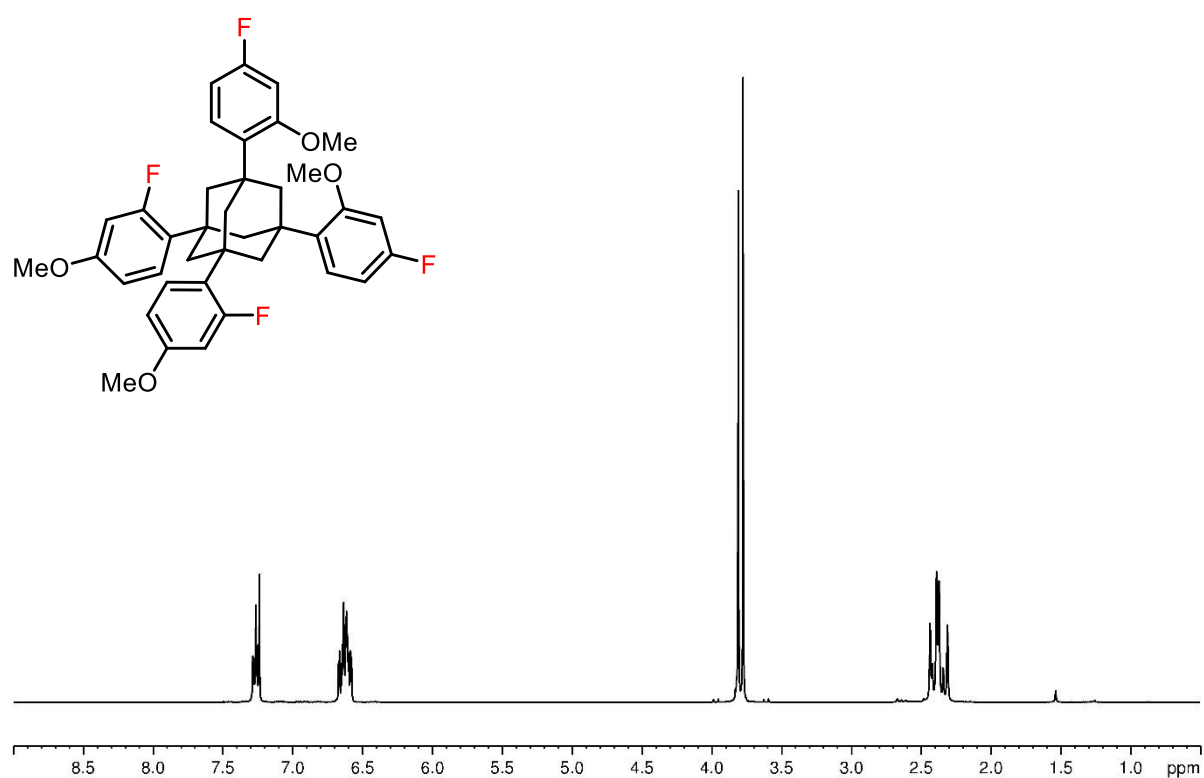
**Figure S30.**  $^1\text{H}$ -NMR spectrum of compound 22 ( $\text{CDCl}_3$ , 400 MHz).



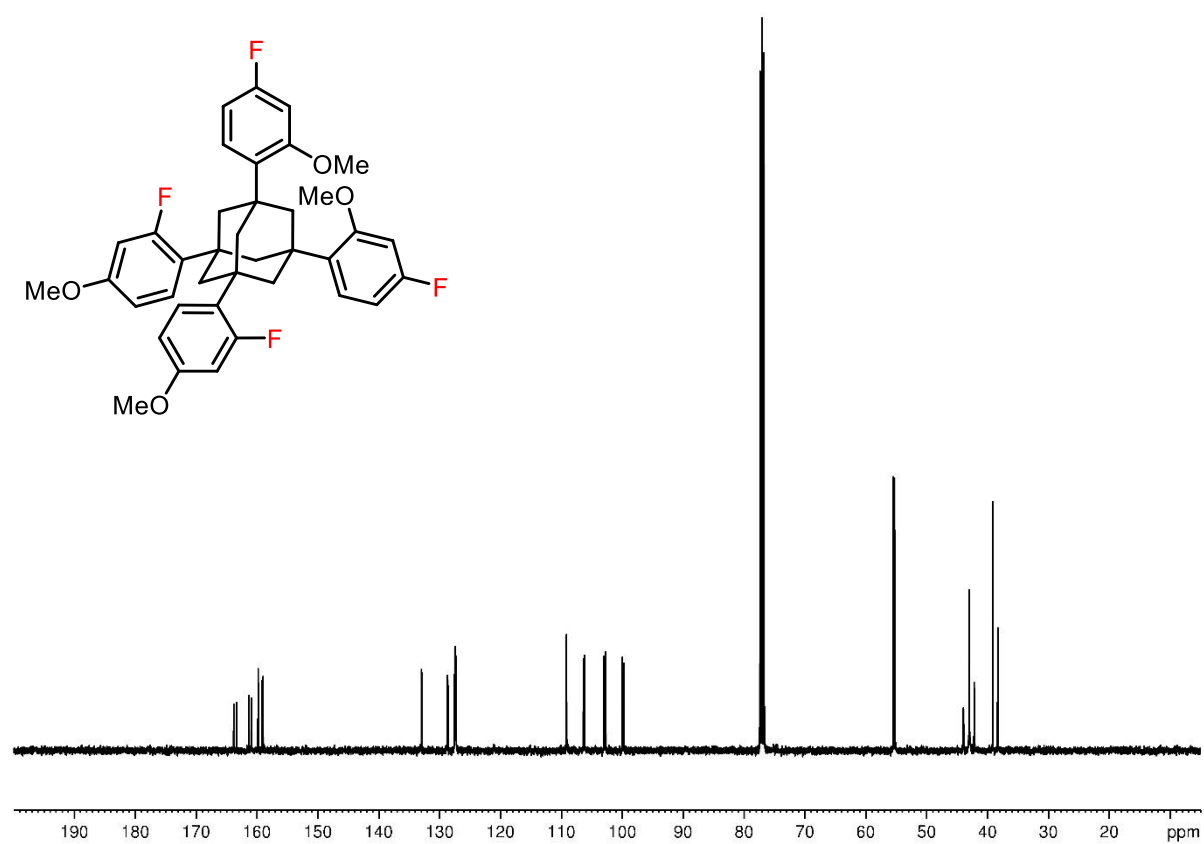
**Figure S31.** <sup>13</sup>C-NMR spectrum of compound **22** (CDCl<sub>3</sub>, 101 MHz).



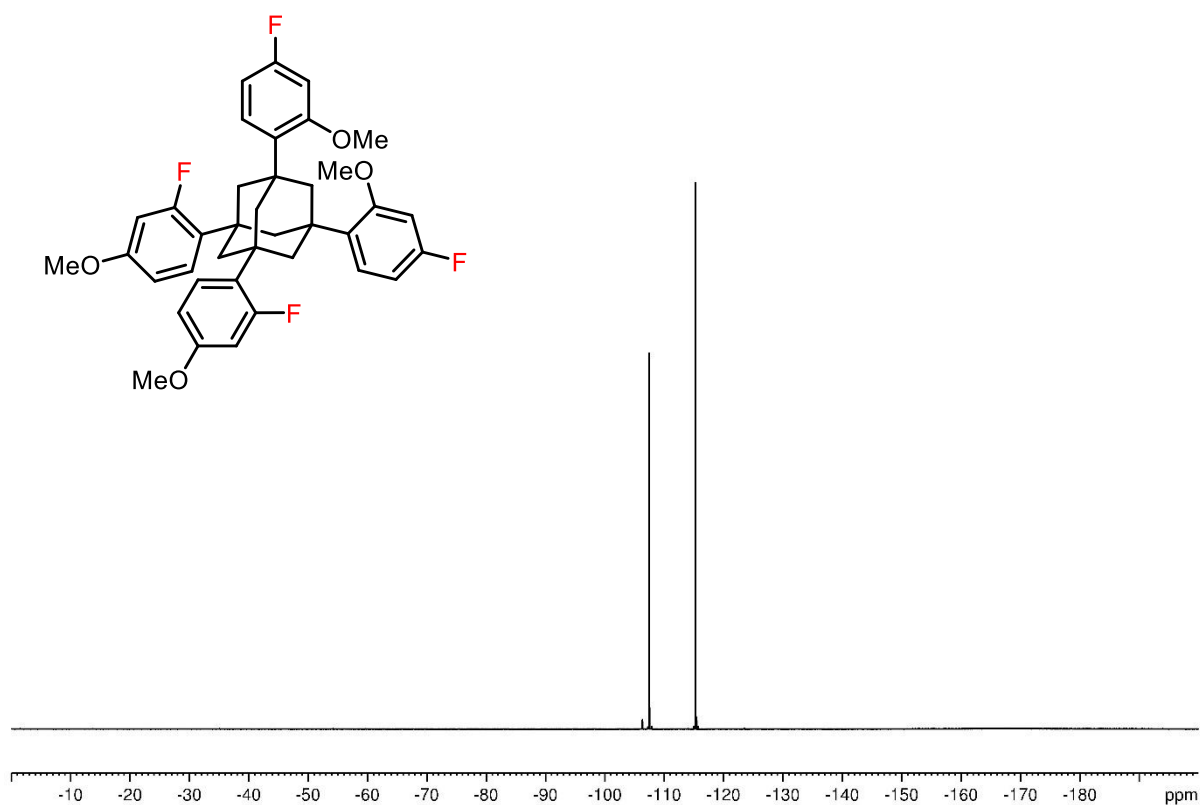
**Figure S32.** <sup>19</sup>F-NMR spectrum of compound **22** (CDCl<sub>3</sub>, 376 MHz).



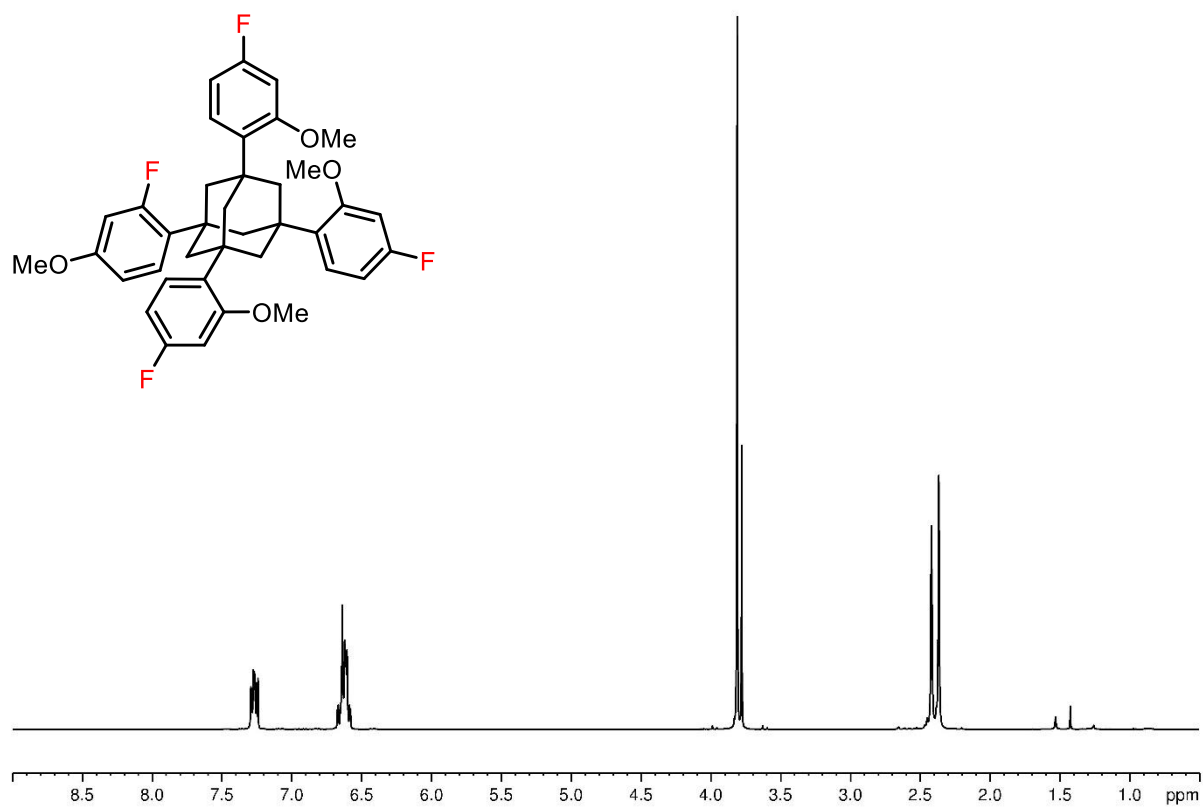
**Figure S33.** <sup>1</sup>H-NMR spectrum of compound **23** (CDCl<sub>3</sub>, 400 MHz).



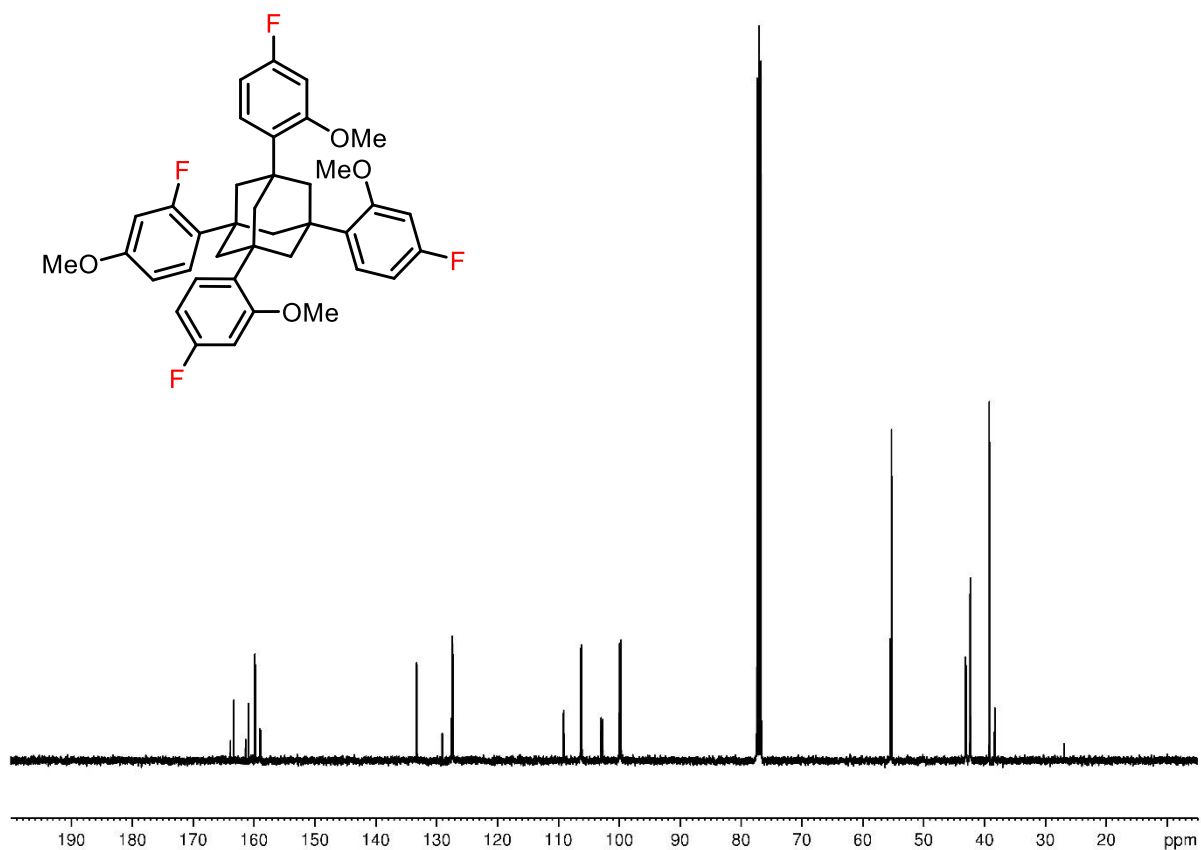
**Figure S34.** <sup>13</sup>C-NMR spectrum of compound **23** (CDCl<sub>3</sub>, 101 MHz).



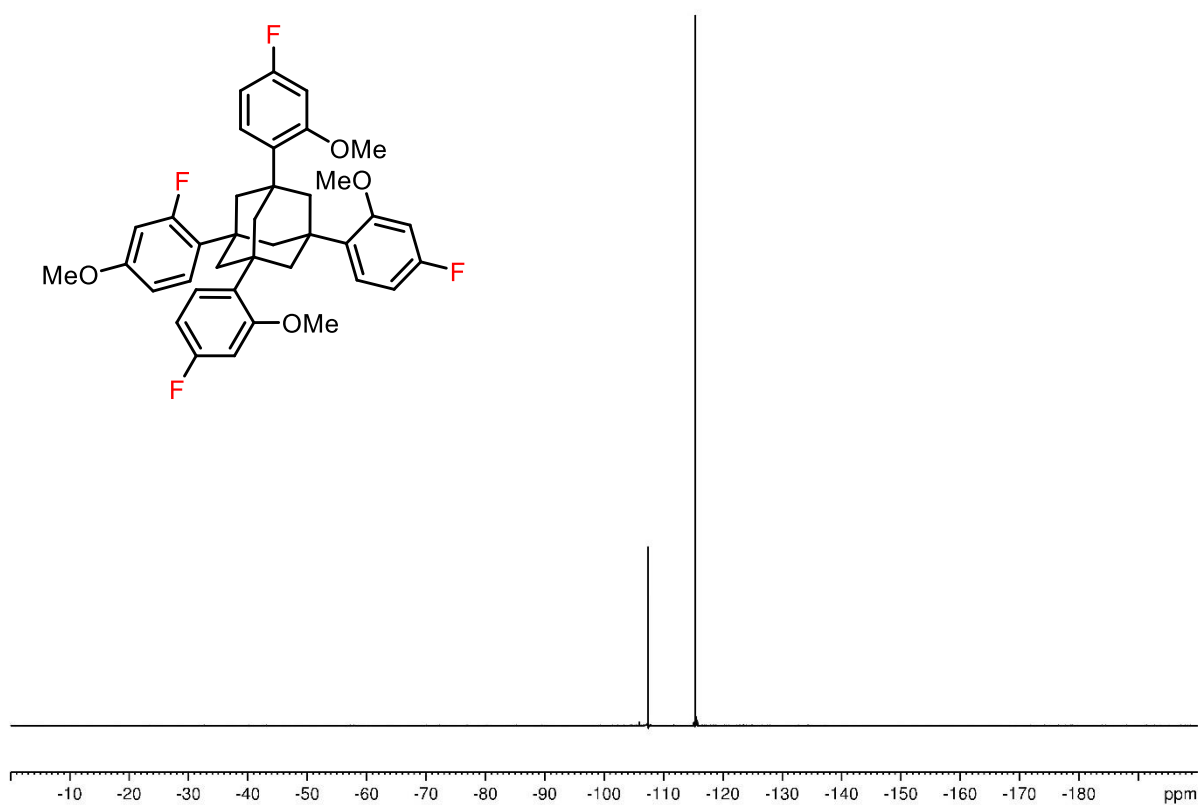
**Figure S35.**  $^{19}\text{F}$ -NMR spectrum of compound **23** ( $\text{CDCl}_3$ , 376 MHz).



**Figure S36.**  $^1\text{H}$ -NMR spectrum of compound **23** ( $\text{CDCl}_3$ , 400 MHz).



**Figure S37.**  $^{13}\text{C}$ -NMR spectrum of compound **23** ( $\text{CDCl}_3$ , 101 MHz).

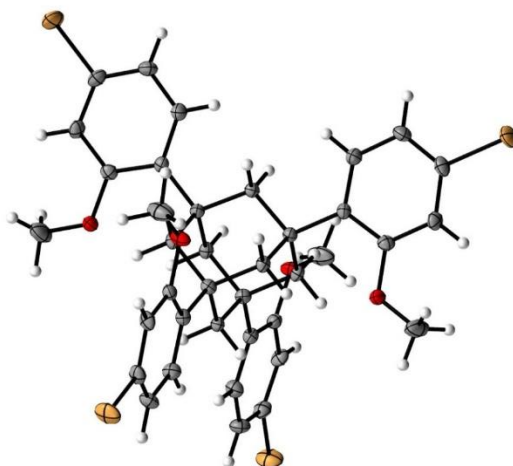


**Figure S38.**  $^{19}\text{F}$ -NMR spectrum of compound **23** ( $\text{CDCl}_3$ , 376 MHz).

## 2. X-ray Crystal Structures

### Structural details for solvate-free form of iTBro

CCDC deposition number 2251141

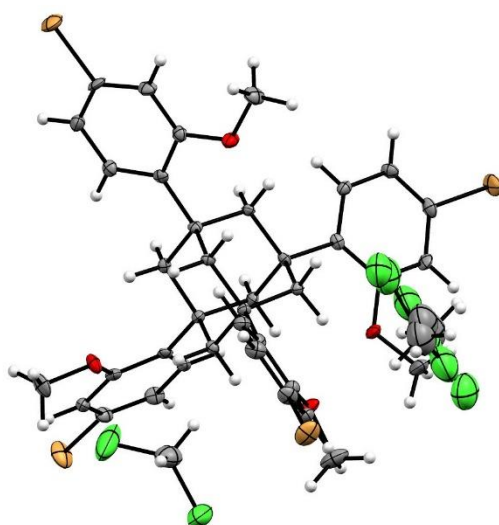


**Figure S39.** One molecule of iTBro in the crystal structure of without inclusion.

Empirical formula	$C_{38}H_{36}Br_4O_4$
Formula weight	876.31
Temperature	140(2) K
Wavelength	0.71073 Å
Crystal system, space group	Tetragonal, $I4(1)/a$
Unit cell dimensions	$a = 19.4378(10)$ Å $\alpha = 90^\circ$ $b = 19.4378(10)$ Å $\beta = 90^\circ$ $c = 9.1102(6)$ Å $\gamma = 90^\circ$
Volume	$3442.1(4)$ Å <sup>3</sup>
Z, Calculated density	4, 1.691 Mg/m <sup>3</sup>
Absorption coefficient	$4.717$ mm <sup>-1</sup>
F(000)	1744
Crystal size	0.240 x 0.081 x 0.048 mm
Theta range for data collection	2.095 to 26.454 °
Limiting indices	$-24 \leq h \leq 24, -24 \leq k \leq 24, -11 \leq l \leq 11$
Reflections collected / unique	24082 / 1777 [R(int) = 0.0651]
Completeness to theta = 25.242	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.8645 and 0.4337
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	1777 / 0 / 106
Goodness-of-fit on F <sup>2</sup>	1.007
Final R indices [I > 2σ(I)]	R1 = 0.0314, wR2 = 0.0634
R indices (all data)	R1 = 0.0557, wR2 = 0.0674
Extinction coefficient	n/a
Largest diff. peak and hole	0.377 and -0.271 e. Å <sup>-3</sup>

## Structural details of iTBro with encapsulated dichloromethane

CCDC deposition number 2251142



**Figure S40.** Structure of iTBro and two molecules of encapsulated dichloromethane

Empirical formula	$C_{39.50} H_{39} Br_4 Cl_3 O_4$
Chemical formula moiety	$C_{38} H_{36} Br_4 O_4, 1.5 CH_2 Cl_2$
Formula weight	1003.69
Temperature	140(2) K
Wavelength	0.71073 Å
Crystal system, space group	Triclinic, P-1
Unit cell dimensions	$a = 12.5470(9)$ Å $\alpha = 80.520(4)^\circ$ $b = 12.5556(10)$ Å $\beta = 77.892(4)^\circ$ $c = 13.9319(12)$ Å $\gamma = 66.382(3)^\circ$
Volume	$1958.2(3)$ Å <sup>3</sup>
Z, Calculated density	2, 1.702 Mg/m <sup>3</sup>
Absorption coefficient	$4.355$ mm <sup>-1</sup>
F(000)	998
Crystal size	0.310 x 0.242 x 0.150 mm
Theta range for data collection	1.777 to 26.440 °
Limiting indices	$-15 \leq h \leq 15, -14 \leq k \leq 15, -17 \leq l \leq 17$
Reflections collected / unique	32061 / 8011 [R(int) = 0.0436]
Completeness to theta = 25.242	99.9 %
Absorption correction	Numerical
Max. and min. transmission	0.6239 and 0.4113
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	8011 / 102 / 501
Goodness-of-fit on F <sup>2</sup>	1.026
Final R indices [I > 2sigma(I)]	R1 = 0.0413, wR2 = 0.0773
R indices (all data)	R1 = 0.0701, wR2 = 0.0856
Extinction coefficient	0.0024(2)
Largest diff. peak and hole	2.051 and -1.961 e. Å <sup>-3</sup>