

Supporting Information
for DOI: 10.1055/a-0042-1753061

© 2022. Thieme. All rights reserved.

Georg Thieme Verlag KG, Rüdigerstraße 14, 70469 Stuttgart, Germany

Supporting Information

Synthesis of β -arylbutenolides mediated by $\text{BF}_3 \cdot \text{OME}_2$

Beatriz Finêncio Miorin, Fernanda Amorim Santos, Rosangela da Silva de Laurentiz *

Universidade Estadual Paulista (Unesp), Faculdade de Engenharia de Ilha Solteira,
Departamento de Física e Química, Ilha Solteira/SP, Brazil

Table of Contents

Table of Contents

Materials and Instrumentation	S2
General Procedure for the synthesis β -arylbutenolides derivatives	S2
Spectral Data of Compounds	S3
NMR (^1H and ^{13}C) and HRMS Spectra of compounds 3a–3d	S6

MATERIALS AND INSTRUMENTATION

Tetronic acid was purchased from Alfa Aesar and the other chemicals and solvents were purchased from Sigma-Aldrich and were used as received without further purification. Microwave irradiation was carried out in a Reactor Discover Reflux (CEM corporation, 300 W) in open vessel. All experiments were monitored by thin layer chromatography using Merck silica gel 60 F₂₅₄ plates. The products were visualized with 356 and 254 nm UV lamps. The ¹H and ¹³C NMR spectra were determined on a Bruker DRX 400 spectrometer in DMSO-d₆. Melting points were measured with a Buchi B-540 melting point apparatus and are uncorrected. The high-resolution mass spectral data were obtained using a Bruker Daltonics - micrOTOF-Q, fitted with an ESI operating in the positive ion mode.

General Procedure for the synthesis of β-arylbutenolides derivatives

A mixture of tetronic acid (1.0 mmol), phenol (1.0 mmol) or aryl ether, and BF₃.OMe₂ (2.0 eq) in CHCl₃ (3 mL) was taken in a reaction flask equipped with a small magnetic stirring bar and the reflux condenser. The mixture was irradiated in a microwave reactor at a power of 220 W. The reaction was monitored by TLC, and after consumption of the starting materials was cooled to room temperature. The solvent was removed from the vacuum and the precipitated crude product was washed with a mixture of hexane-ethyl acetate (8:2). When sesamol was used as phenol, the mixture was maintained at room temperature under stirring for 24 h. The proceeding after the reaction was the same described above.

Spectral Data of Compounds

4-(2-hydroxy-4,5-dimethoxyphenyl)furan-2(5H)-one (3a)

Yield: 92% (0.2149 g). Mp 236-237 °C.

¹H NMR (DMSO-d₆, 400 MHz): δ 10.3 (s, OH), 6.94 (s, 1H), 6.46 (s, 1H), 6.40 (s, 1H), 5.21 (s, 2H), 3.67 (s, 3H), 3.64 (s, 3H).

¹³C NMR (DMSO-d₆, 125 MHz): δ 174.4, 162.6, 152.8, 152.7, 142.1, 111.3, 109.5, 107.5, 100.4, 72.4, 56.3, 55.4.

HRMS (ESI⁺): m/z [M+H]⁺ calcd for C₁₂H₁₂O₅Na: 259.0581; found: 259.0572.

4-(6-hydroxybenzo[d][1,3]dioxol-5-yl)furan-2(5H)-one (3b)

Yield: 85% (0.1885g). Mp 228-230 °C.

¹H NMR (DMSO-d₆, 400 MHz): δ 10.63 (s, OH), 7.14 (s, 1H), 6.54 (s, 1H), 6.46 (s, 1H), 6.03 (s, 2H), 5.26 (s, 2H).

¹³C NMR (DMSO-d₆, 125 MHz): δ 174.4, 162.5, 154.0, 150.8, 140.7, 109.8, 108.3, 106.4, 101.7, 97.6, 72.5.

HRMS (ESI⁺): m/z [M+H]⁺ calcd for C₁₁H₈O₅Na: 243.0268; found: 243.0263.

4-(2,4-dihydroxyphenyl)furan-2(5H)-one (3c)

Yield: 95% (0.1825 g). Mp 241-242 °C.

¹H NMR (DMSO-d₆, 400 MHz): δ 10.56 (s, OH), 10.09 (s, OH), 7.34 (d, 1H, J = 8.6 Hz), 6.40 (d, 1H, J = 2.6 Hz), 6.37 (d, 1H, J = 1.3 Hz), 6.33 (dd, 1H, J = 2.2 Hz and J = 8.6 Hz), 5.26 (brs, 2H).

¹³C NMR (DMSO-d₆, 125 MHz): δ 174.6, 162.4, 161.6, 159.0, 130.0, 109.8, 108.8, 107.8, 102.6, 72.5.

HRMS (ESI⁺): m/z [M+H]⁺ calcd for C₁₀H₈O₄Na: 215.0319; found: 215.0313.

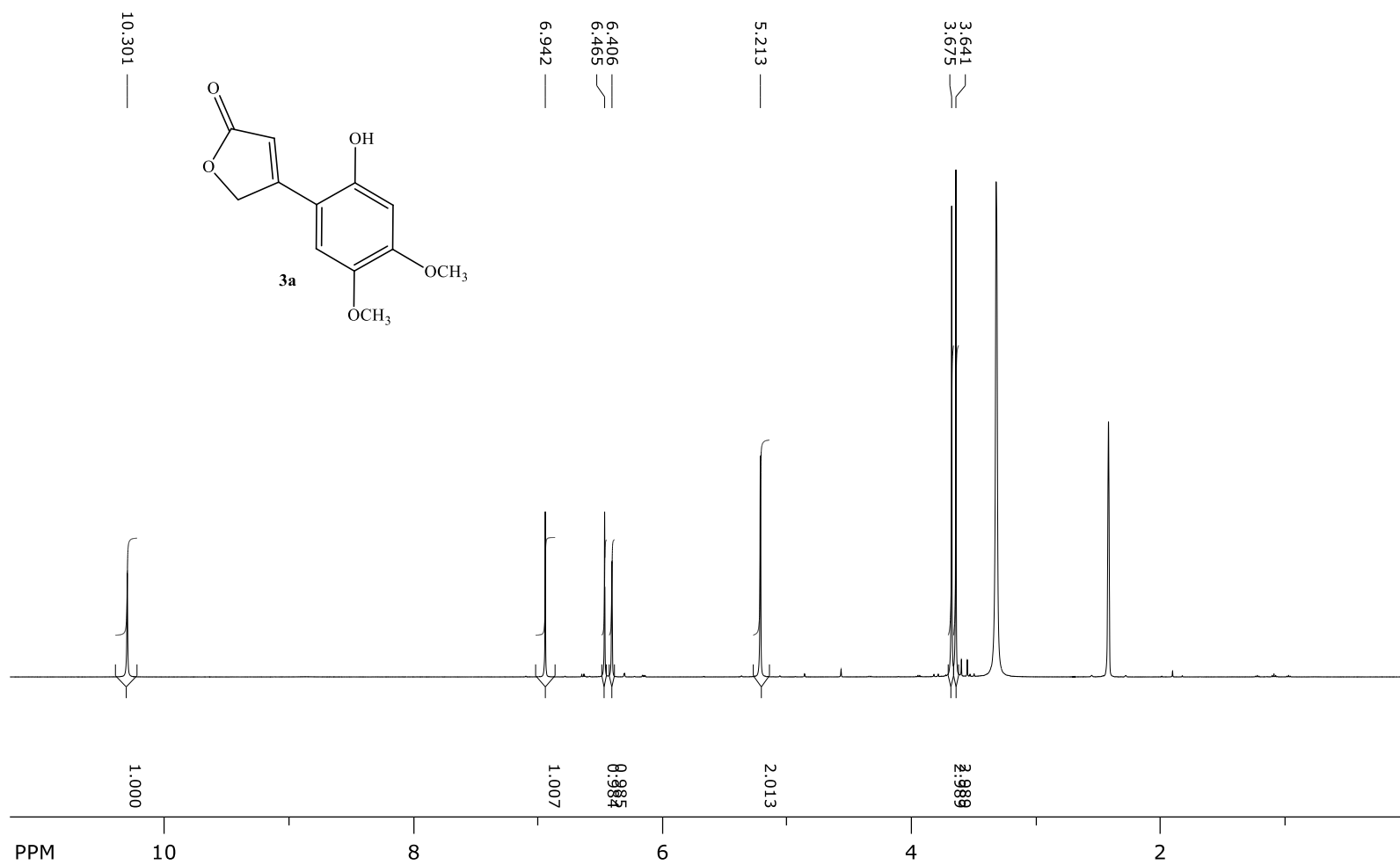
4-(2,4-dimethoxyphenyl)furan-2(5H)-one (3d)

White solid Mp 205-207 °C. Yield: 0.0202 g (92%)

¹H NMR (DMSO-d₆, 400 MHz): δ 7.52 (d, 1H, J = 8.66 Hz), 6.69 (d, 1H, J = 2.33 Hz), 6.64 (dd, 1 H, J = 8.65 Hz, J = 2.33 Hz), 6.50-6.46 (m, 1H), 5.29 (d, 2H, J = 1.55 Hz), 3.90 (s, 3H), 3.84 (s, 3H).

¹³C NMR (DMSO-d₆, 125 MHz): δ 174.3, 163.3, 161.1, 160.0, 130.0, 111.2, 111.0, 106.2, 98.6, 72.1, 55.8, 55.6.

HRMS (ESI⁺): m/z [M+Na]⁺ calcd for C₁₂H₁₂O₄Na: 243.0633; found: 243.0632

NMR (^1H and ^{13}C) and HRMS Spectra of compounds 3a –3d**Figure 1S-** ^1H NMR Spectrum of compound 3a in DMSO- d_6

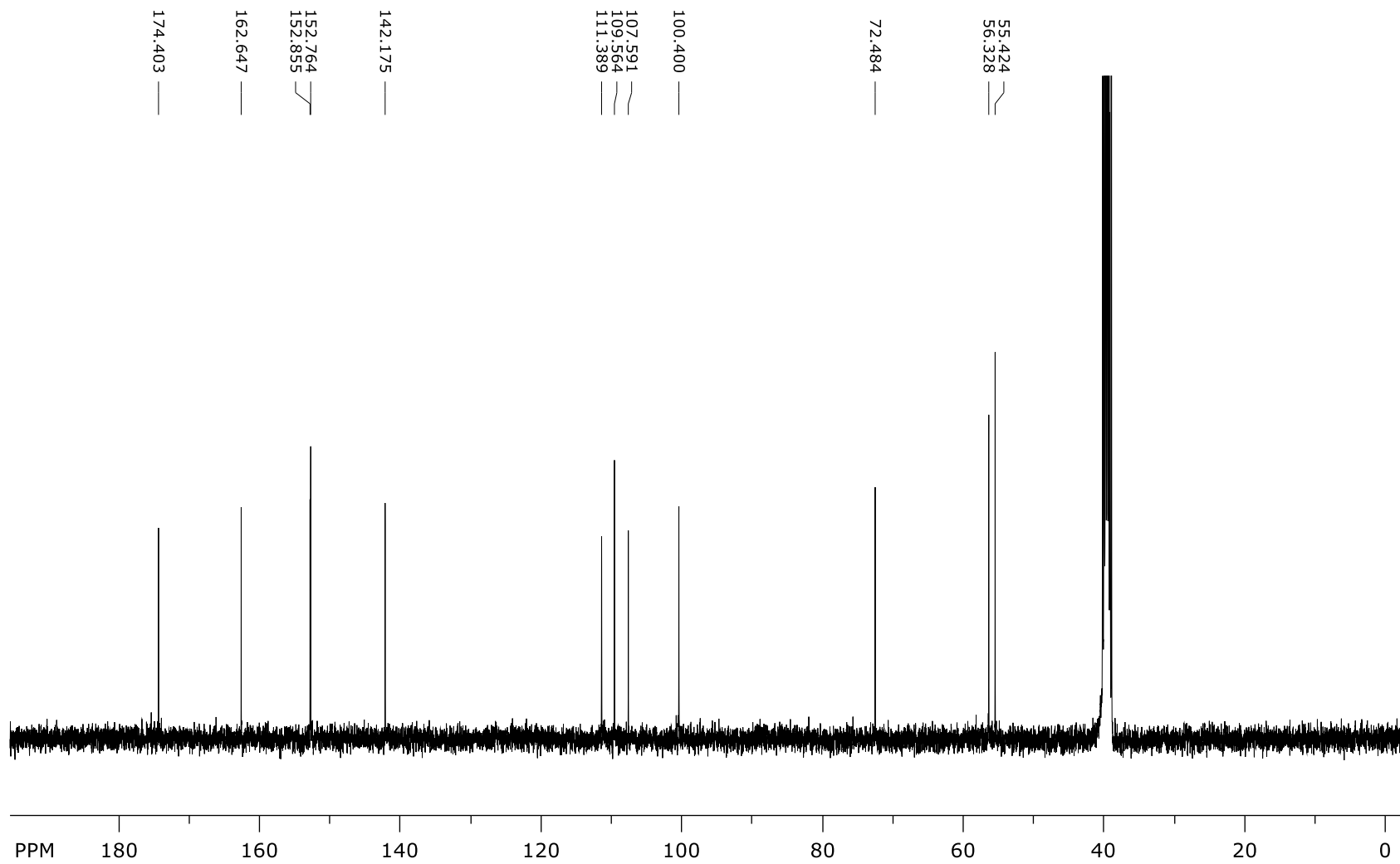


Figure 2S- ¹³C NMR Spectrum of compound **3a** in DMSO-d₆

Analysis Info

Analysis Name D:\Data\2022\FAUZEAOUADA\ROSANGELA\27-7-22\rs3a_POS_1-3_01_6309.d
Method INFUSAO POR LC_POS.m
Sample Name rs3a_POS
Comment

Acquisition Date 7/27/2022 2:00:08 PM

Operator BD@AL
Instrument micrOTOF

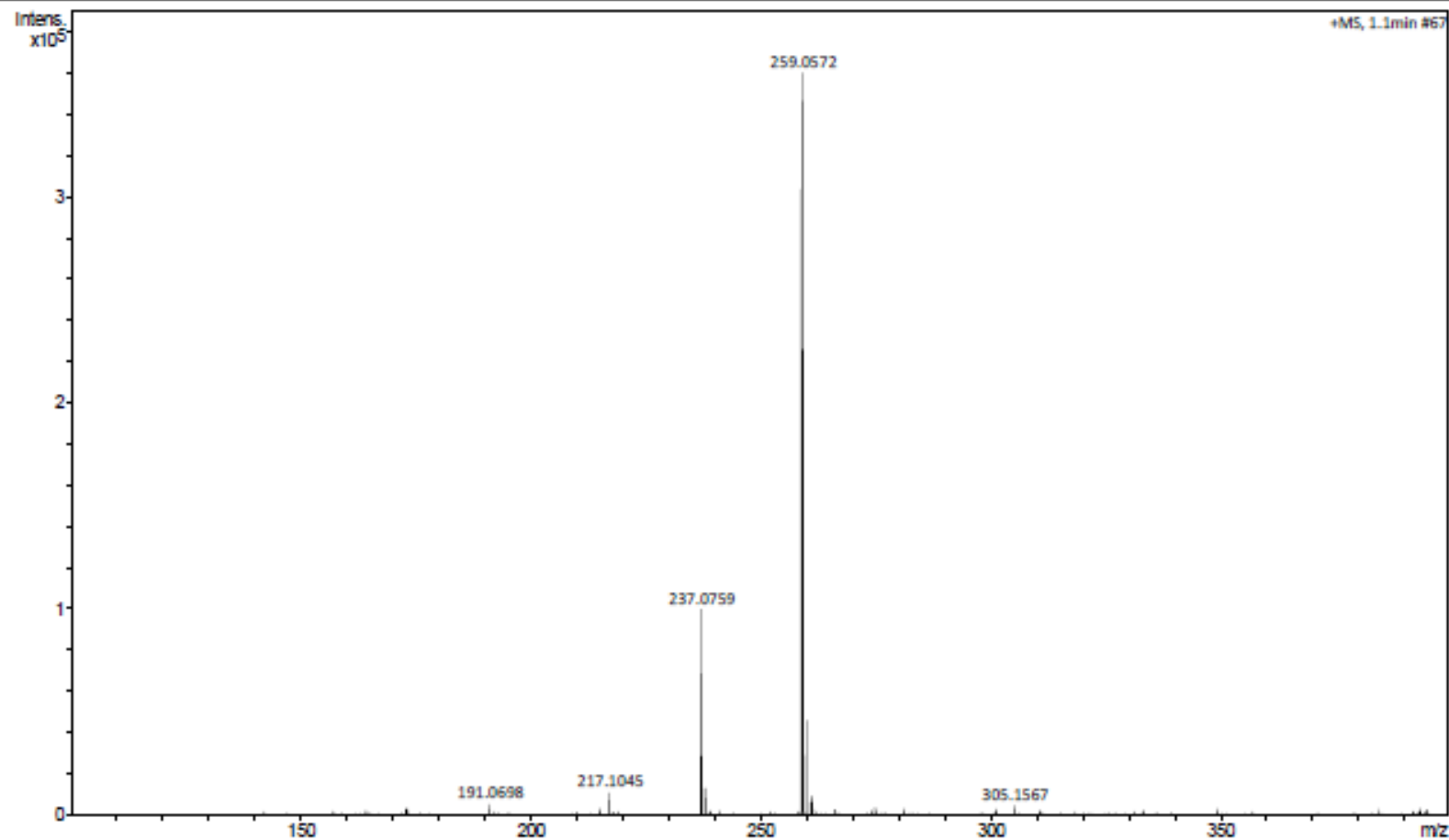


Figure 3S- HRMS spectrum of compound 3a

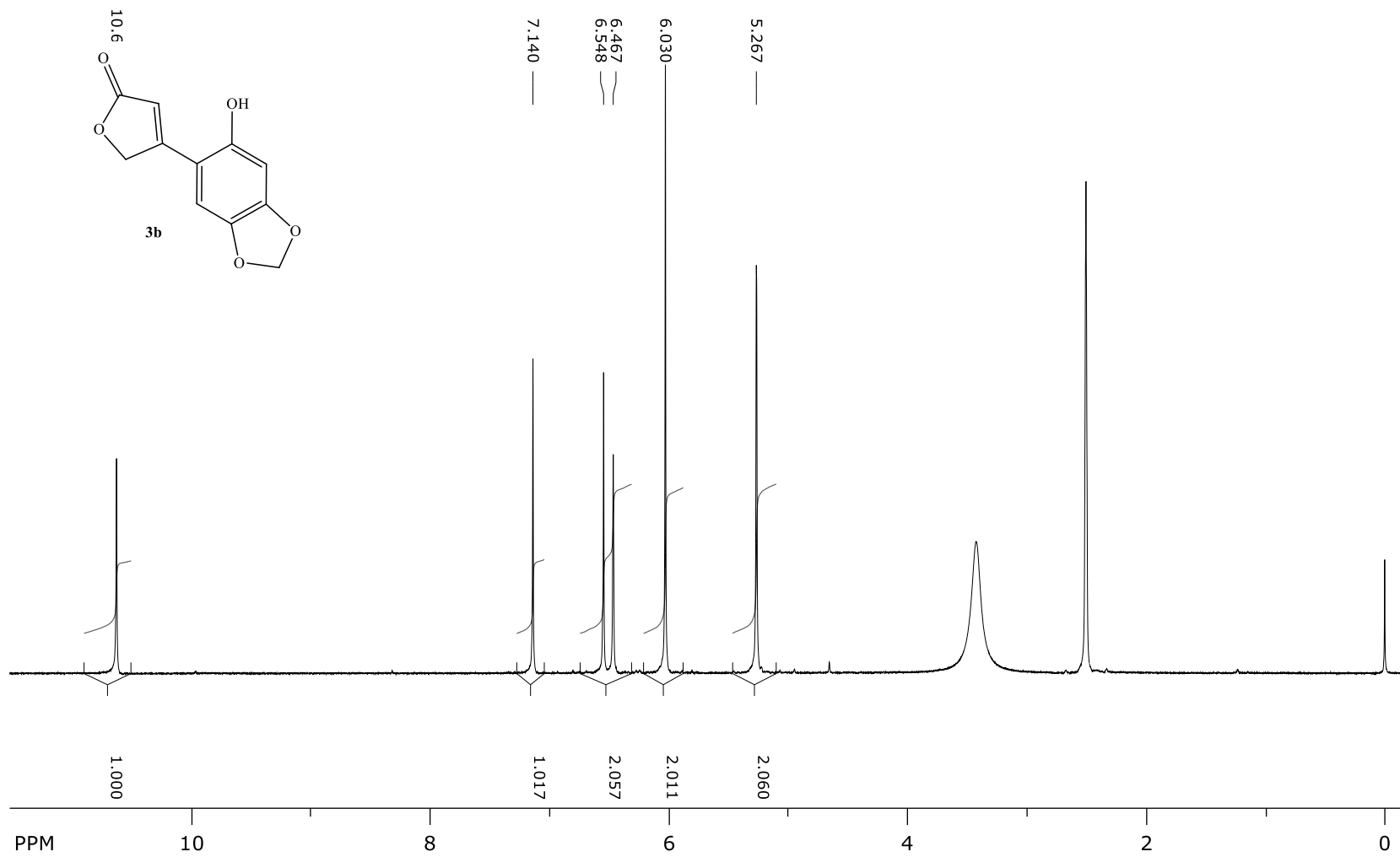


Figure 4S- ¹H NMR Spectrum of compound **3b** in DMSO-d₆

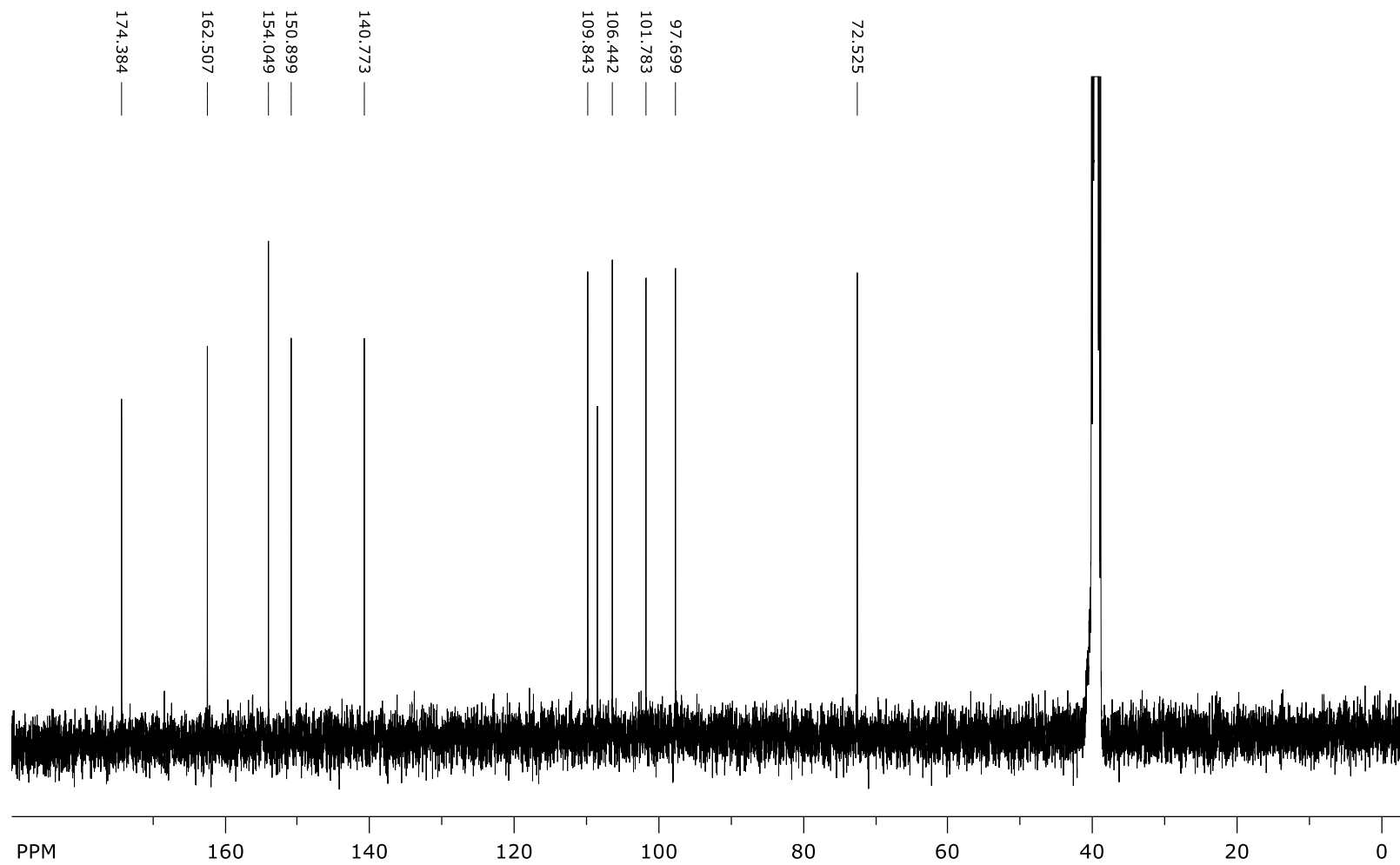


Figure 5S- ^{13}C NMR Spectrum of compound **3b** in DMSO- d_6

Analysis Info

Analysis Name D:\Data\2022\FAUZE AOUADA\ROSANGELA\27-7-22\rs3b_POS_1-4_01_6310.d
Method INFUSAO POR LC_POS.m
Sample Name rs3b_POS
Comment

Acquisition Date 7/27/2022 2:10:51 PM

Operator BD@AL
Instrument micrOTOF

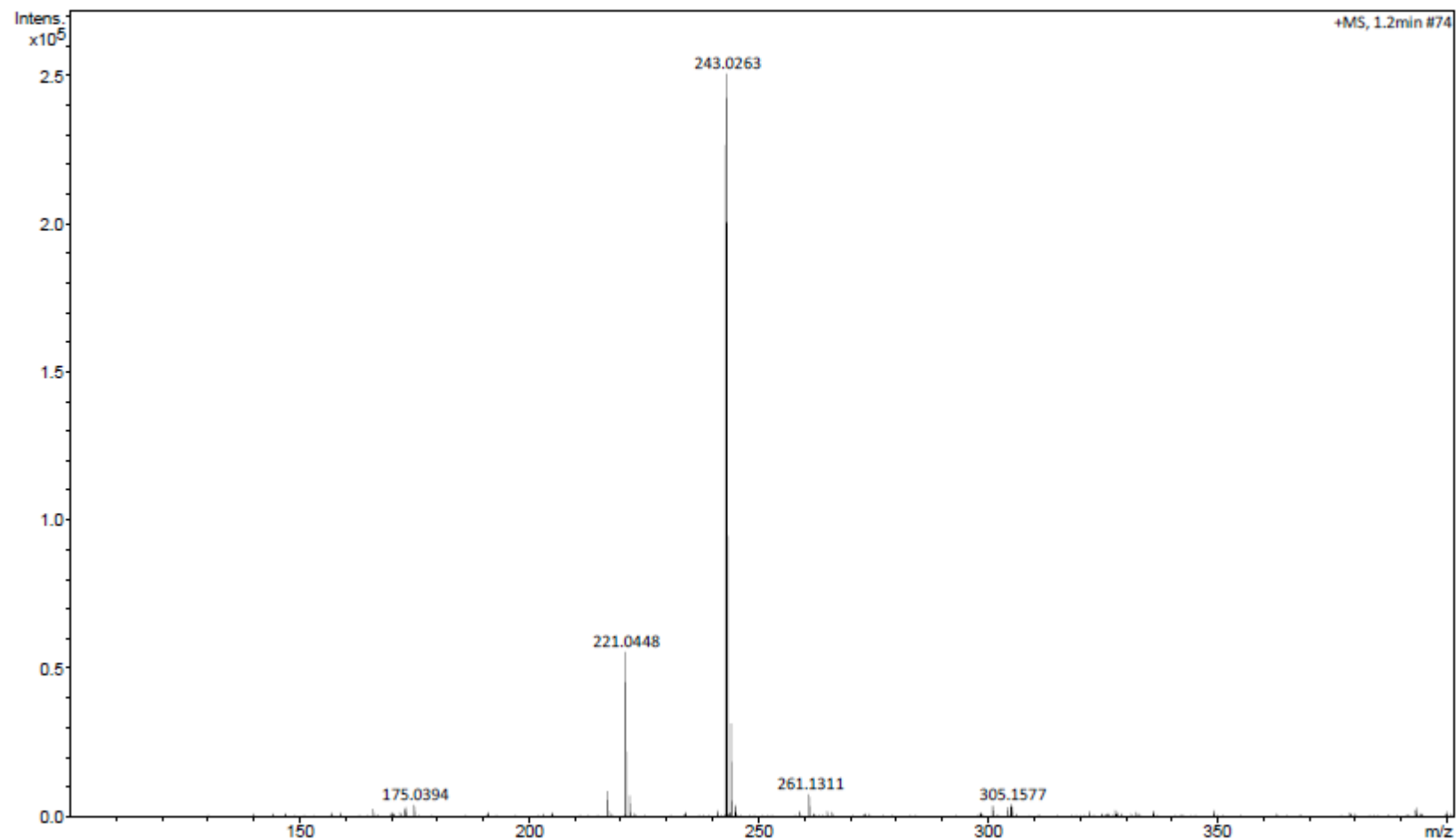


Figure 6S. HRMS spectrum of compound **3b**

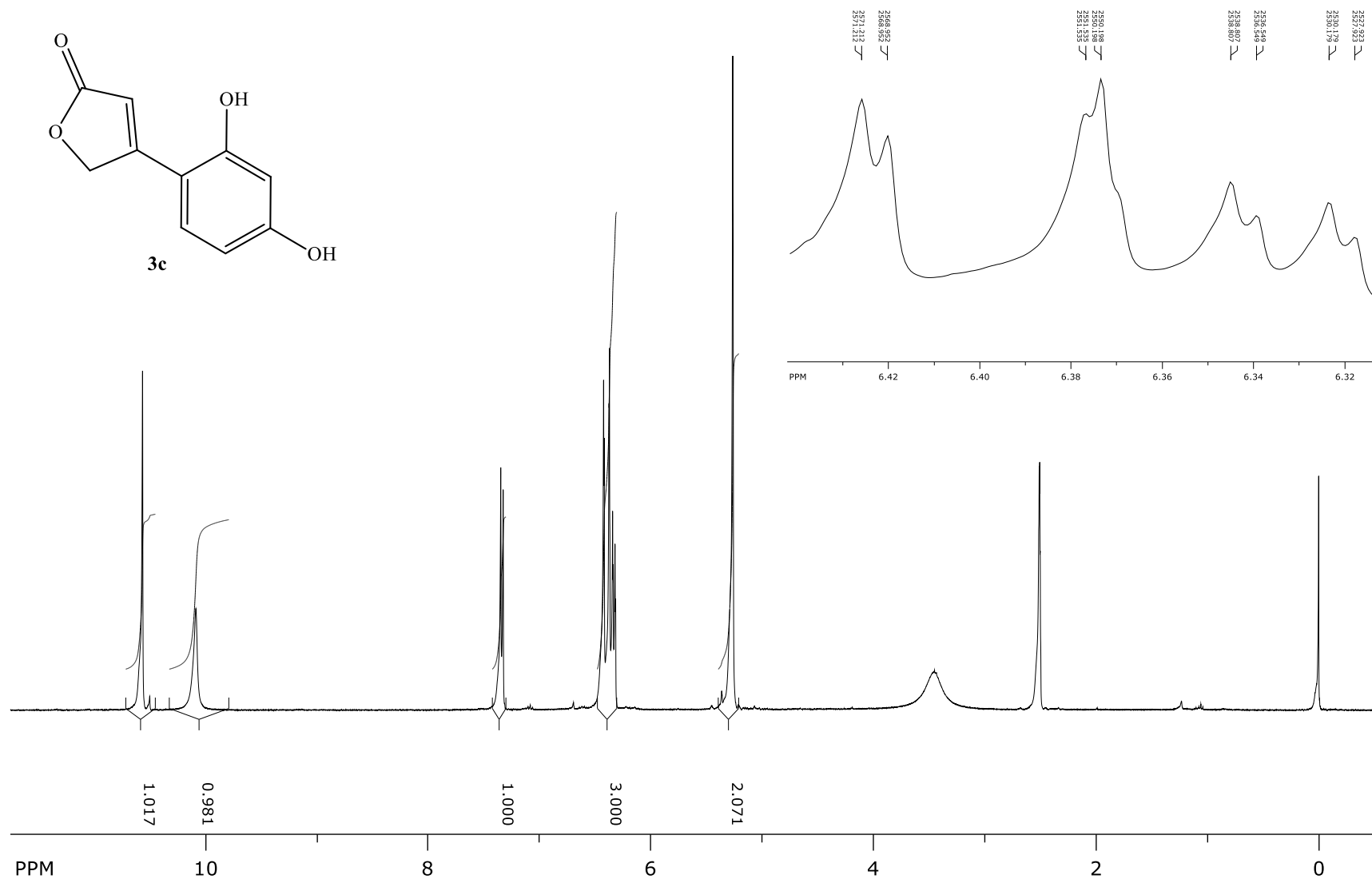


Figure 7S- ¹H NMR Spectrum of compound **3c** in DMSO-d₆

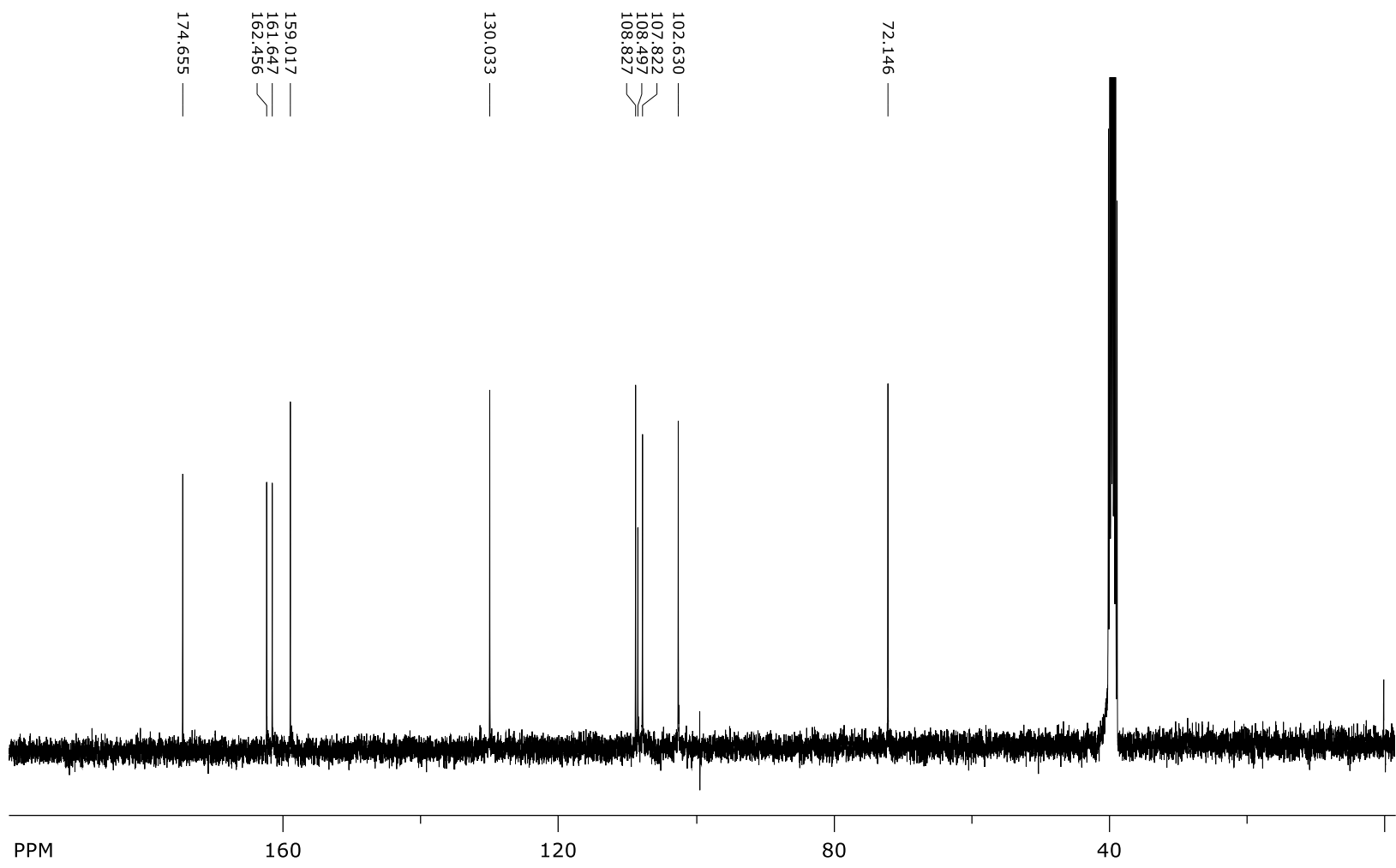


Figure 8S- ¹³C NMR Spectrum of compound 3c in DMSO-d6

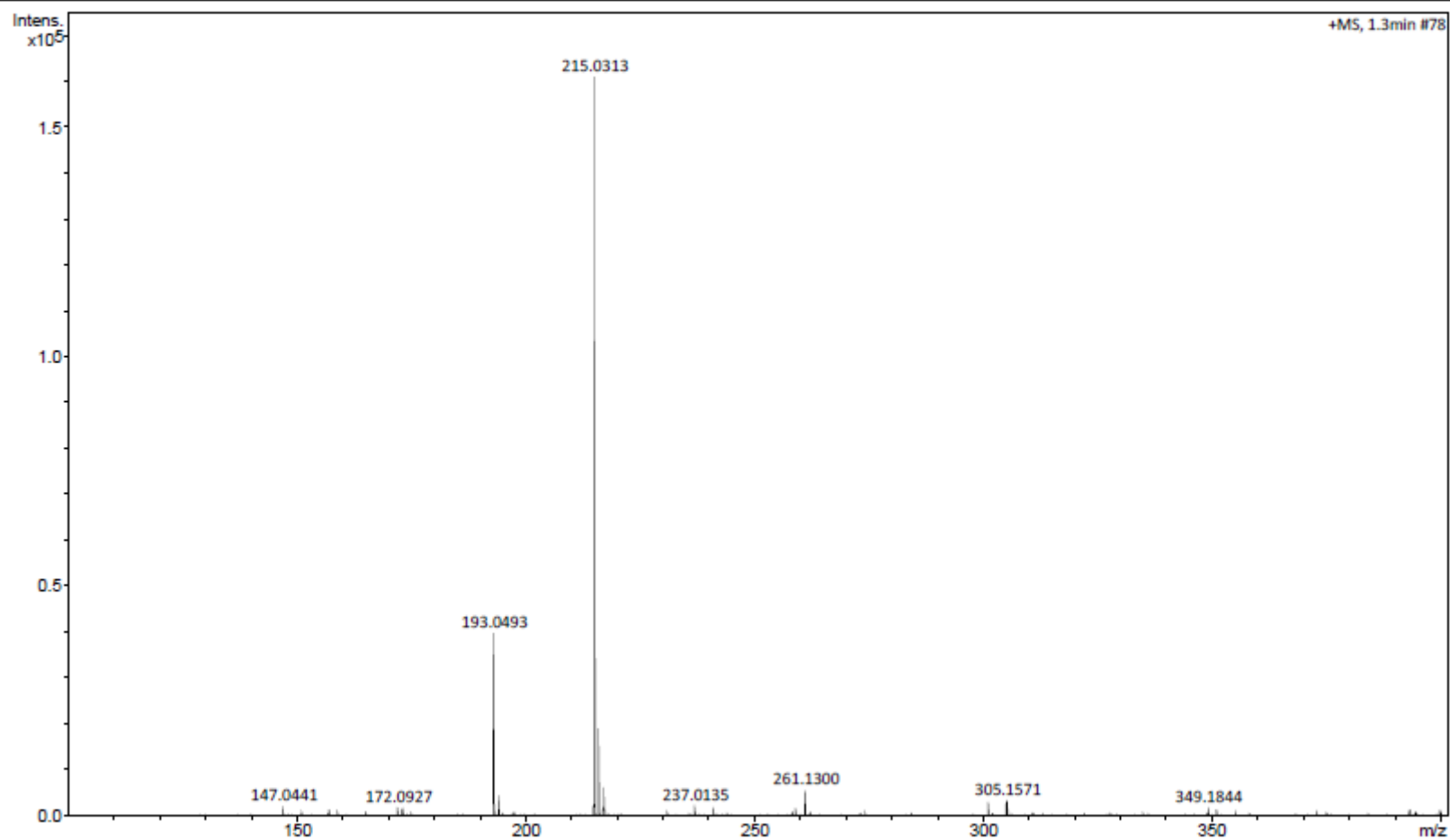
Analysis Info

Analysis Name
Method
Sample Name
Comment

D:\Data\2022\FAUZE AOUADA\ROSANGELA\27-7-22\rs3c_POS_1-5_01_6311.d
INFUSAO POR LC_POS.m
rs3c_POS

Acquisition Date
Operator
Instrument

7/27/2022 2:21:34 PM
BD@AL
micrOTOF

**Figure 9S.** HRMS spectrum of compound **3c**

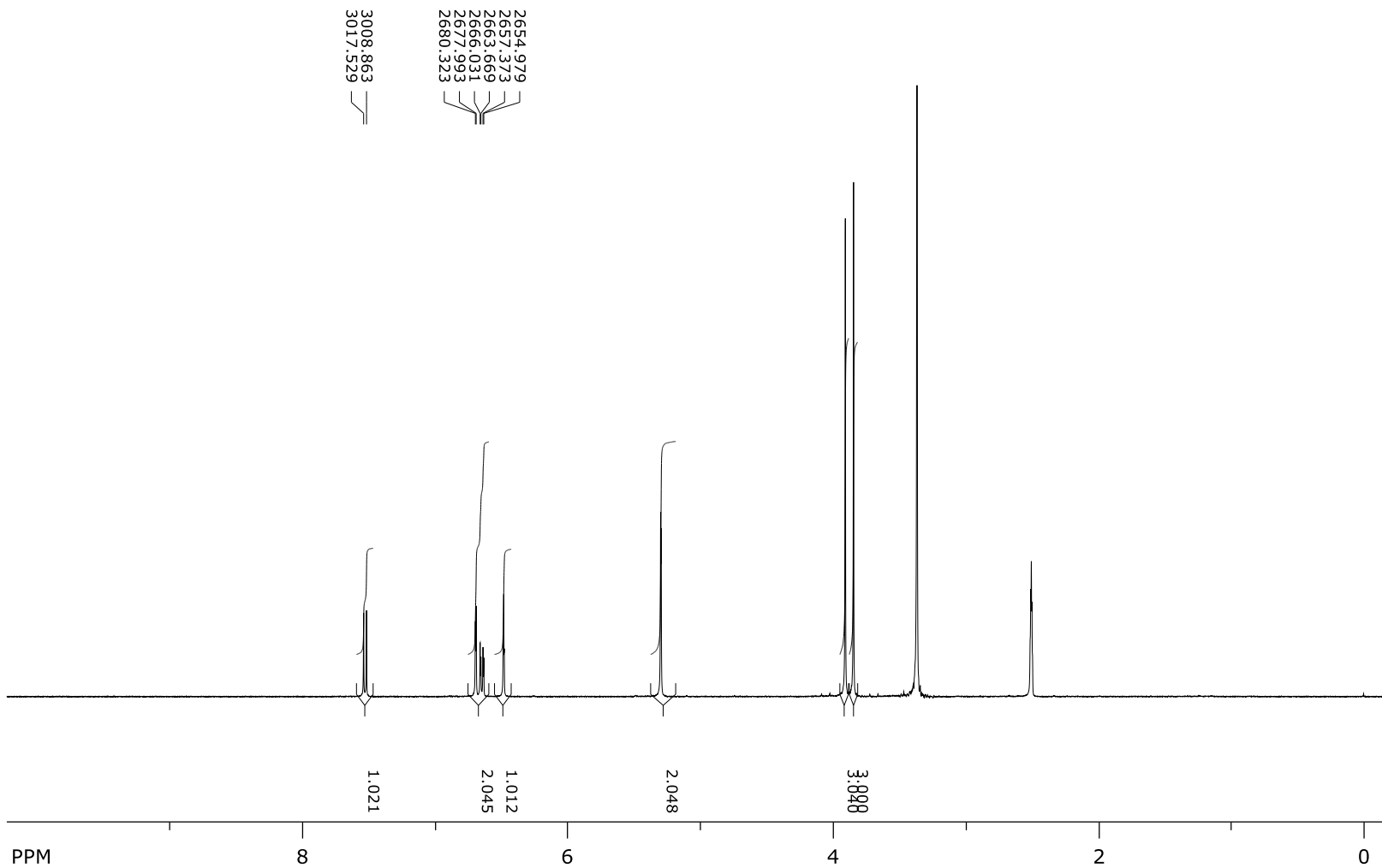


Figure 10S- ^1H NMR Spectrum of compound **3d** in DMSO-d_6

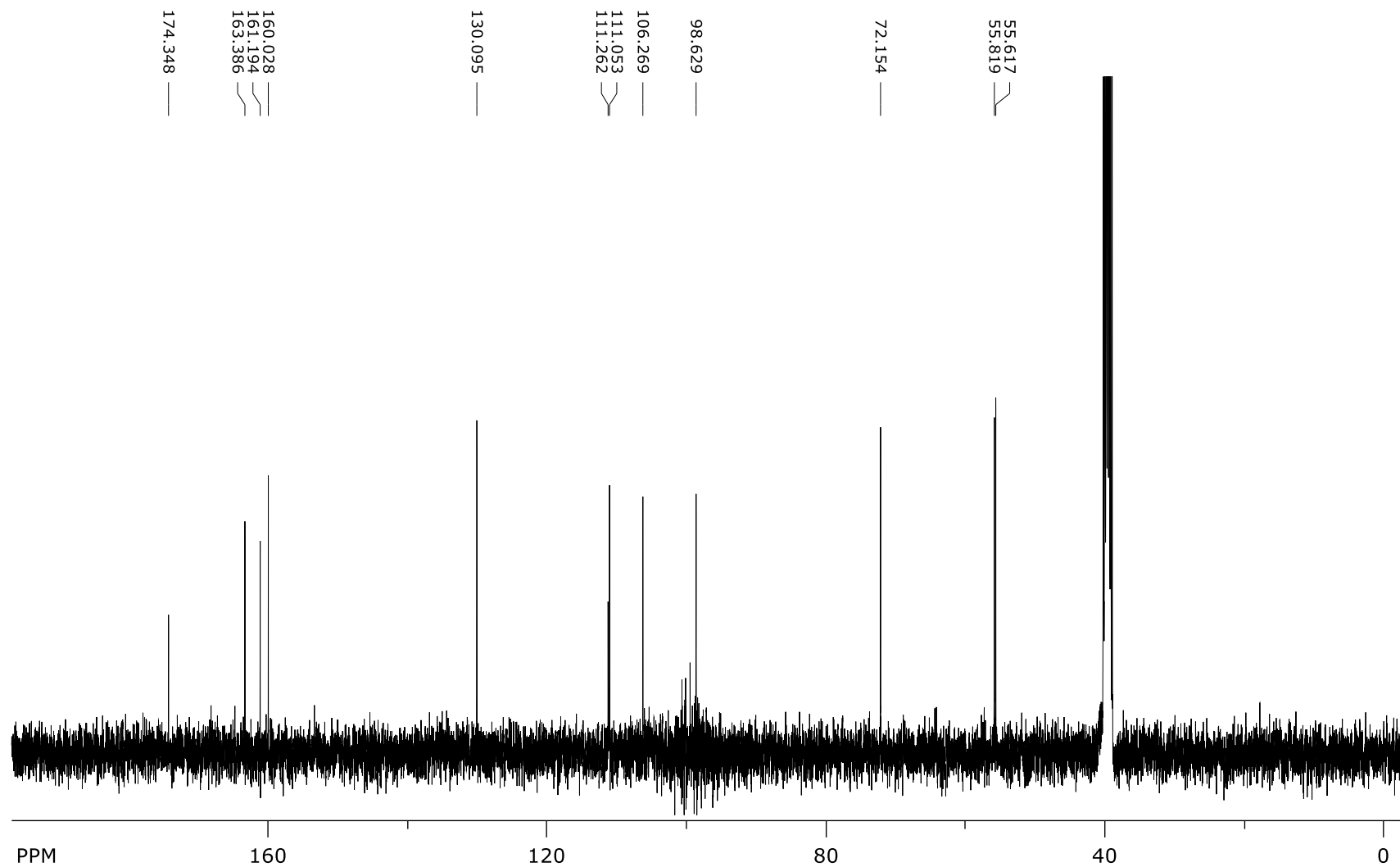


Figure 11S- ¹³C NMR Spectrum of compound **3d** in DMSO-d₆

Analysis Info

Analysis Name D:\Data\2022\FAUZE AOUADA\ROSANGELA\27-7-22\rs3d_POS_1-6_01_6312.d
Method INFUSAO POR LC_POS.m
Sample Name rs3d_POS
Comment

Acquisition Date 7/27/2022 2:32:18 PM

Operator BD@AL
Instrument microTOF

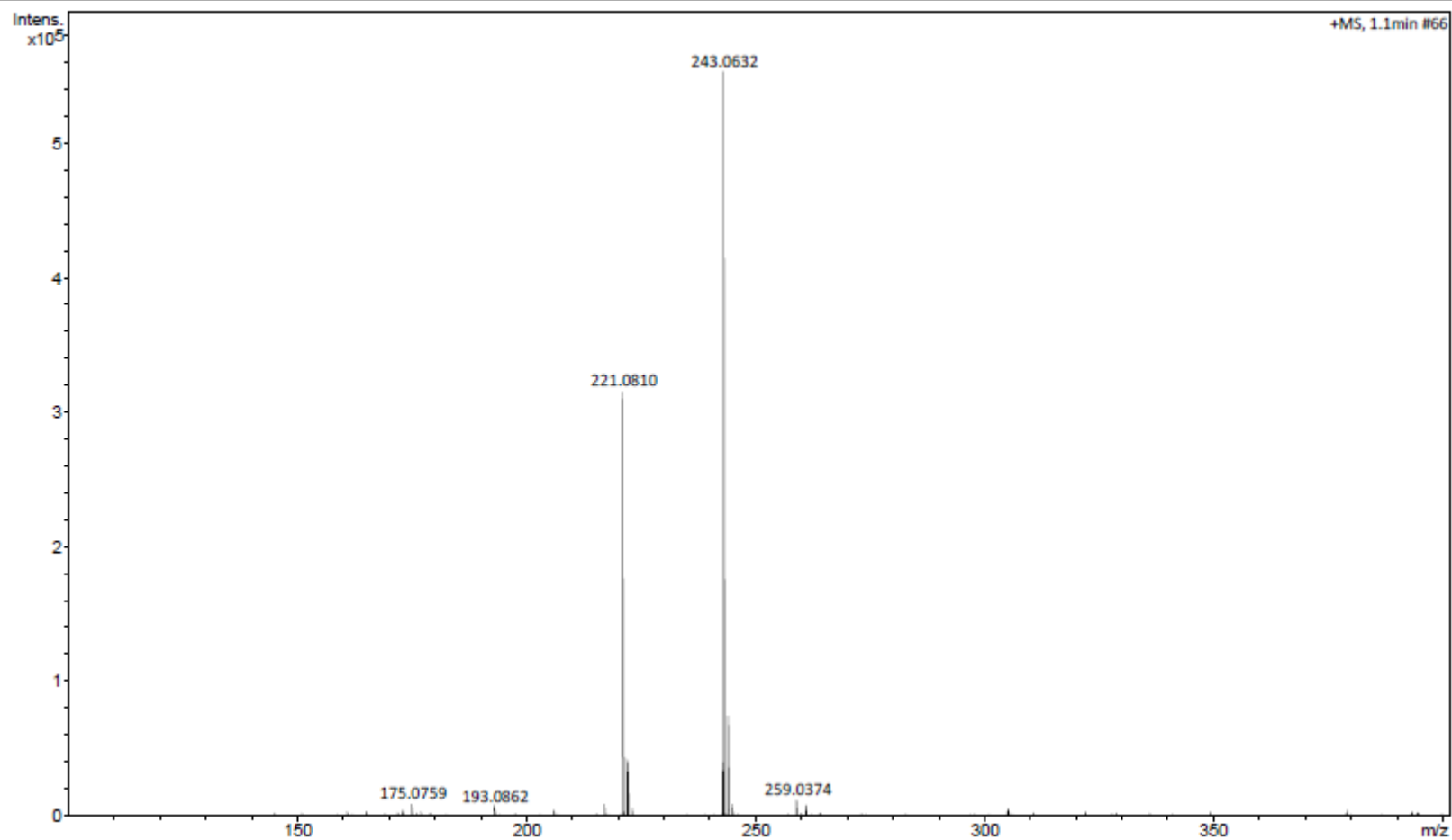


Figure 12S. HRMS spectrum of compound **3d**